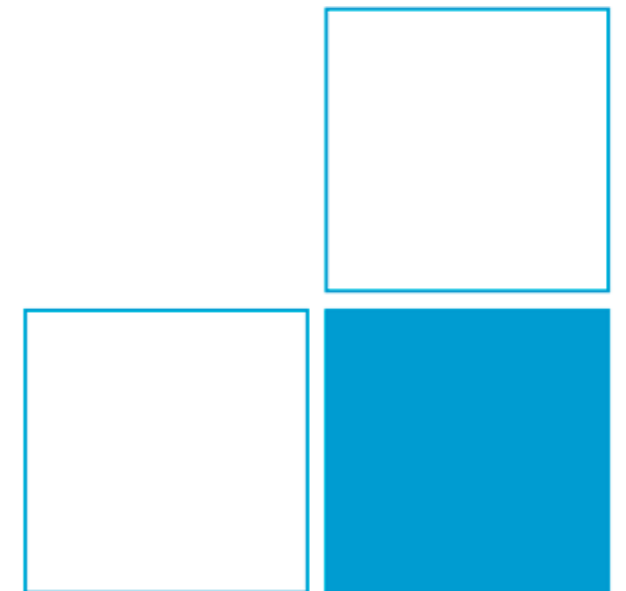


17th CCM meeting May 16/17, 2019

Traceable desorption and outgassing rate measurements

Karl Jousten, PTB, Berlin, Germany

1. Short tutorial on related vacuum physics
2. Method of outgassing rate measurement
3. Written and measurement standards
4. Conclusions



What is desorption, what is outgassing?

Desorption:

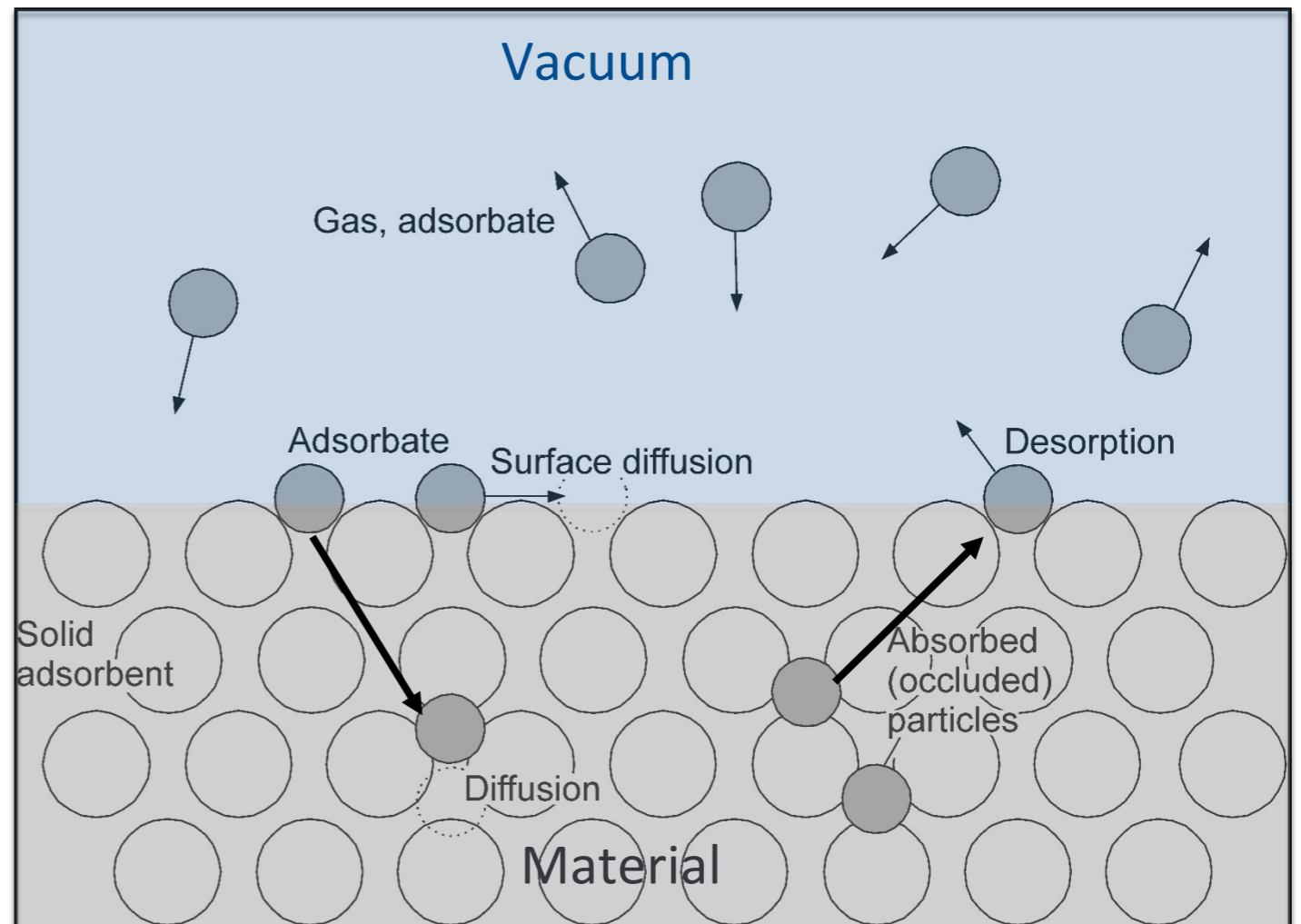
Release of molecules from a surface
– Surface effect

Will be „zero“ after cleaning or
long times – rate depends on time

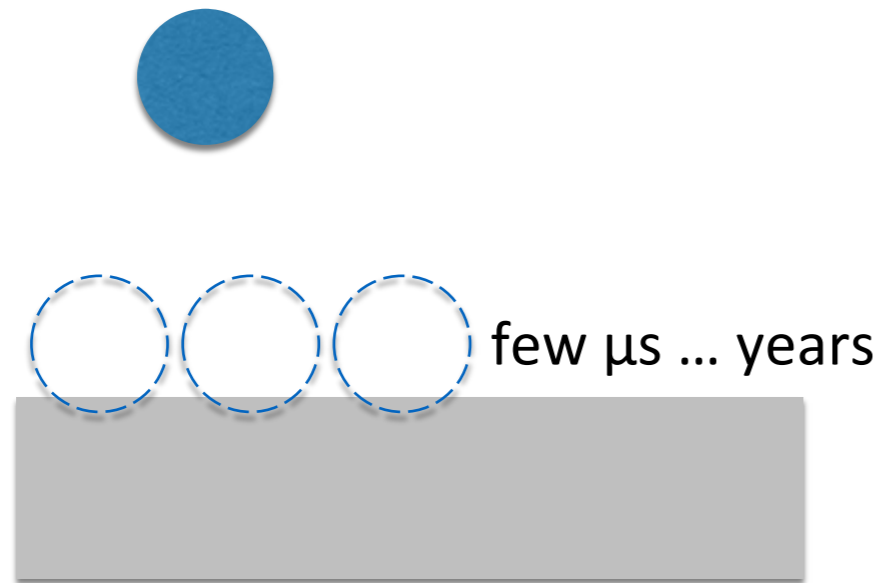
Outgassing:

Release of molecules from a bulk
– Bulk effect

Will „never“ end –
time independent



Desorption rate with time



$\tilde{n}_0(t)$ surface layer

Ideal vacuum/vacuum pump

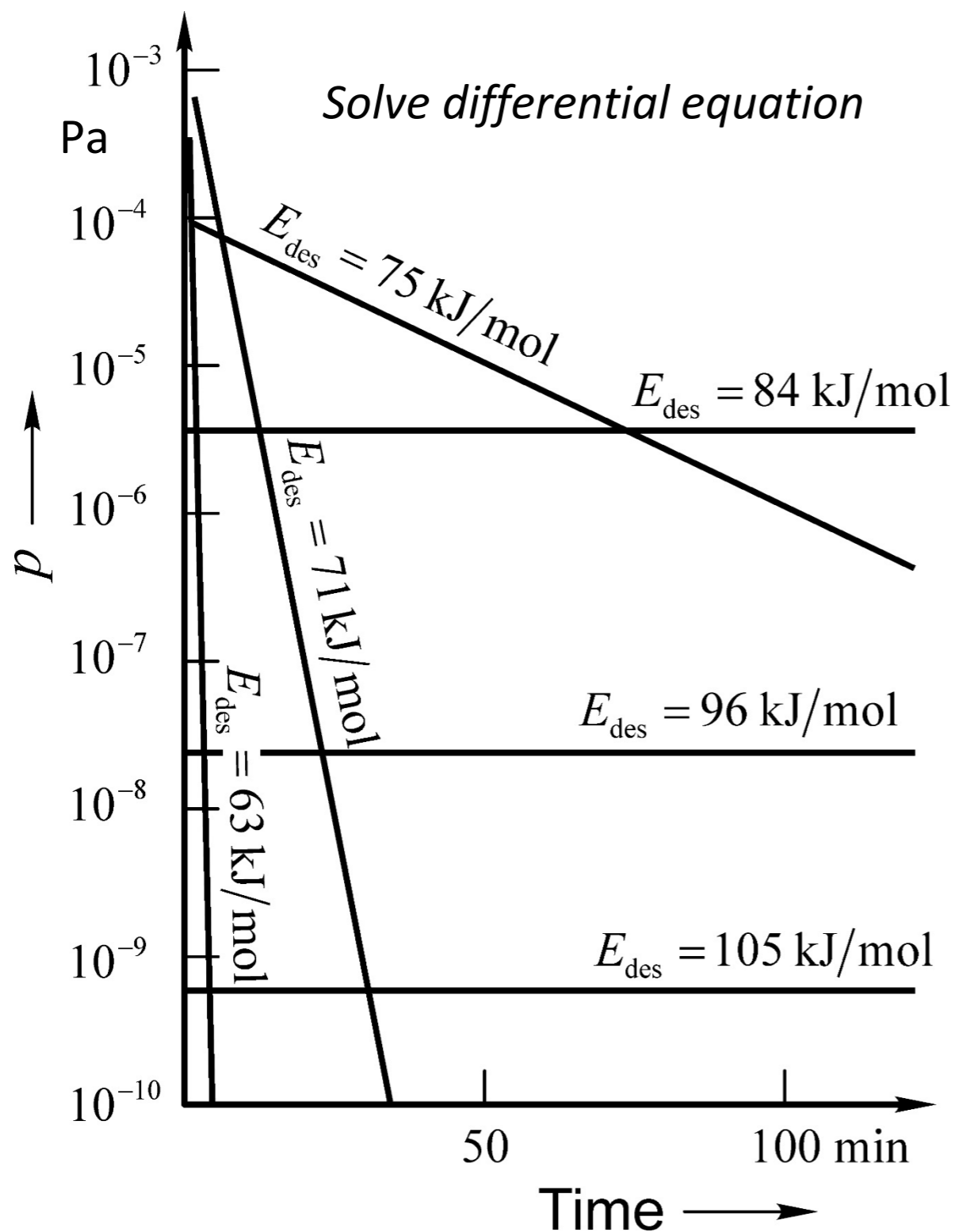
$$\tilde{n}(t) = \tilde{n}_0(t)e^{-t/\tau}$$

$$\tau = \tau_0 e^{E_{\text{des}}/RT}$$

Real case: readsorption

Reduction rate much lower
than exponential

Desorption rate with time at real conditions



H₂O on stainless steel:

E_{des} 80 kJ/mol ... 104 kJ/mol



Similar values for hydrocarbons

Our experience in vacuum technology

In any, not baked-out vacuum system water molecules dominate the residual gas (70 % to 95%).

Typically, any surface exposed to air is “coated” by a few monolayers of H₂O (typically a few up to 30).

All but the last monolayer of H₂O disappear quickly in a vacuum pump down process down to about 1 Pa.

Numbers: Mass per area

Mass/geometrical surface area = Number of sites per area * mass of individual molecule

One monolayer of H₂O molecules on a surface:

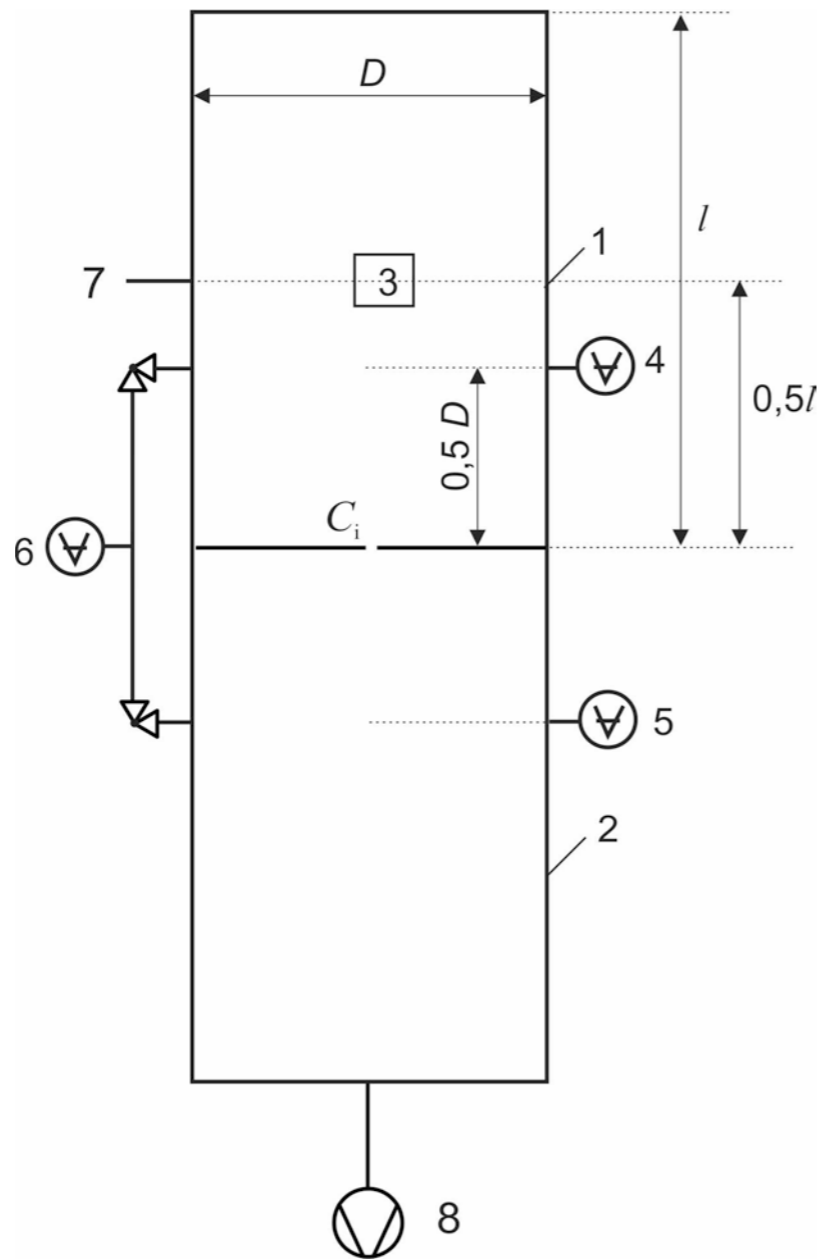
$$10^{15} \text{ cm}^{-2} \cdot 18 \cdot 1.66 \cdot 10^{-27} \text{ kg} = 3 \cdot 10^{-11} \text{ kg cm}^{-2} = 0.03 \text{ } \mu\text{g cm}^{-2}$$

Metrologia 53 (2016), S. Davidson et al., Table 1: “Sorptions value” below 1 Pa:

$$0.013 \text{ } \mu\text{g cm}^{-2} \dots 0.16 \text{ } \mu\text{g cm}^{-2}$$

$$10^{15} \text{ cm}^{-2} = 4.1 \cdot 10^{-3} \text{ Pa L cm}^{-2} \text{ for } 296 \text{ K (preferred and convenient units in vacuum metrology)}$$

How to measure desorption/outgassing rates?



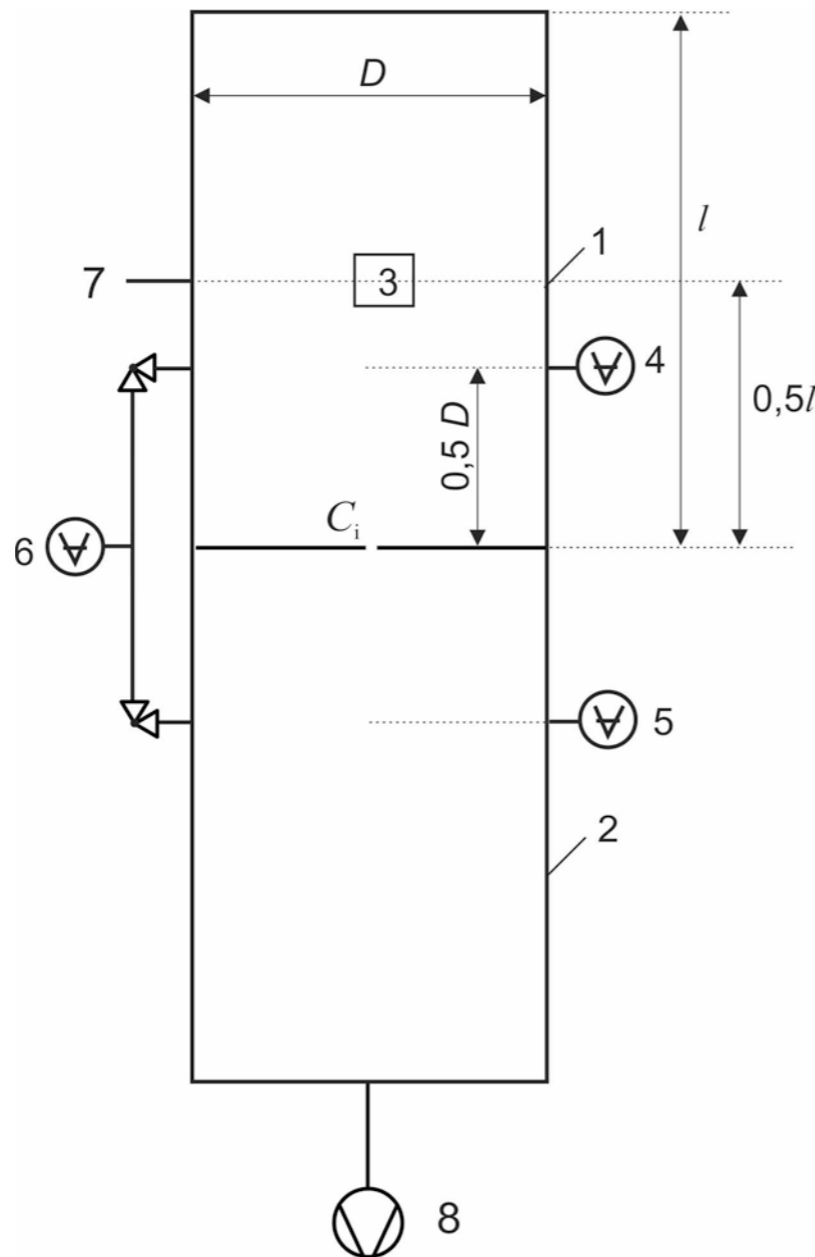
$$q_{\text{out},i} = C_i (p_{1,i} - p_{2,i})$$

time dependent for desorption
time independent for outgassing

ISO TS 20177

Why is it difficult?

$$q_{\text{out},i} = C_i (p_{1,i} - p_{2,i})$$

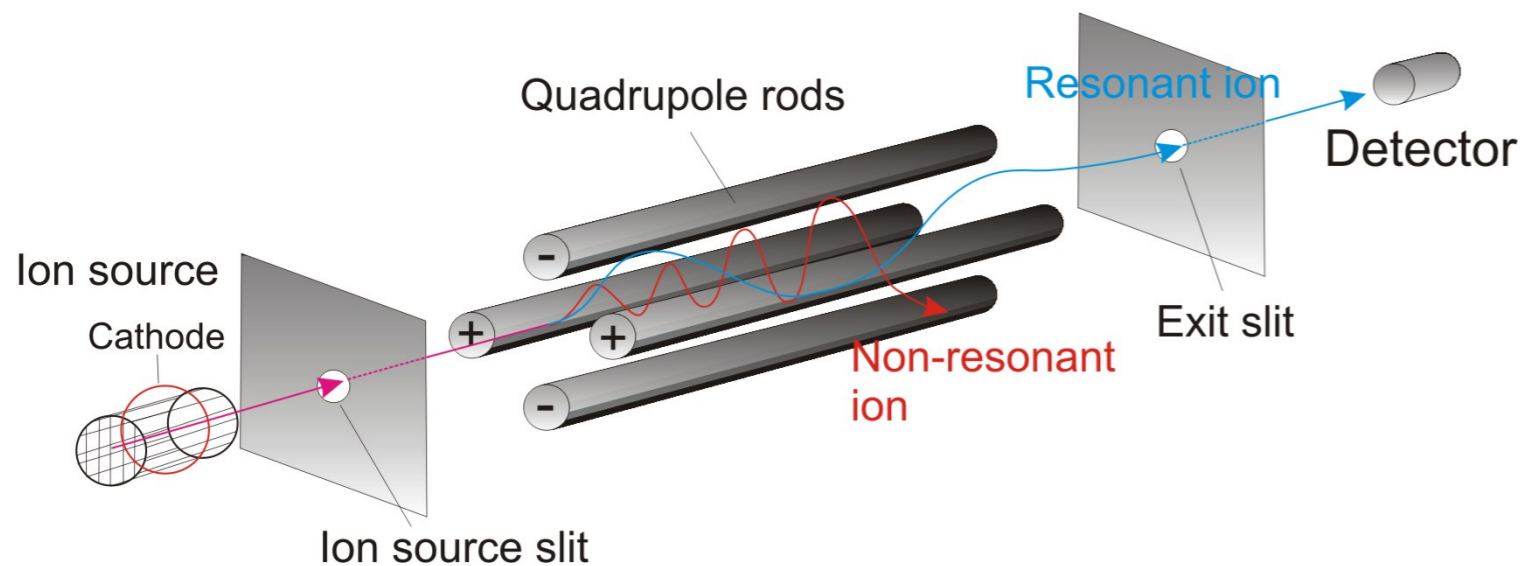


ISO TS 20177

- 1) Pressure gradients in vacuum systems:
 - Positioning of sample
 - Positioning of conductance
 - Positioning of detector
 - Non-infinite pumping speed
- 2) Outgassing/desorption from vacuum chamber: „zero“=residual pressure, its Stability
- 3) Time dependence: When is $t = 0$?
- 4) Mixture of outgassing species: Sensitivity of total pressure vacuum gauge unknown
and the worst:
- 5) Partial pressure measurement: Quadrupole mass spectrometers cannot be calibrated

Quadrupole mass spectrometers

How they work



What we know

- Settings of QMS play an important role: emission current, electron energy, ion energy, field axis potential, m/e resolution, scan speed, multiplier gain
- Settings of each type of QMS have different consequences in their metrological characteristics
- Often settings of individual QMS of the same type have different consequences
- Relative sensitivity for single gas species (to nitrogen) cannot be predicted

Problem of sensitivity ->

Quadrupole mass spectrometers

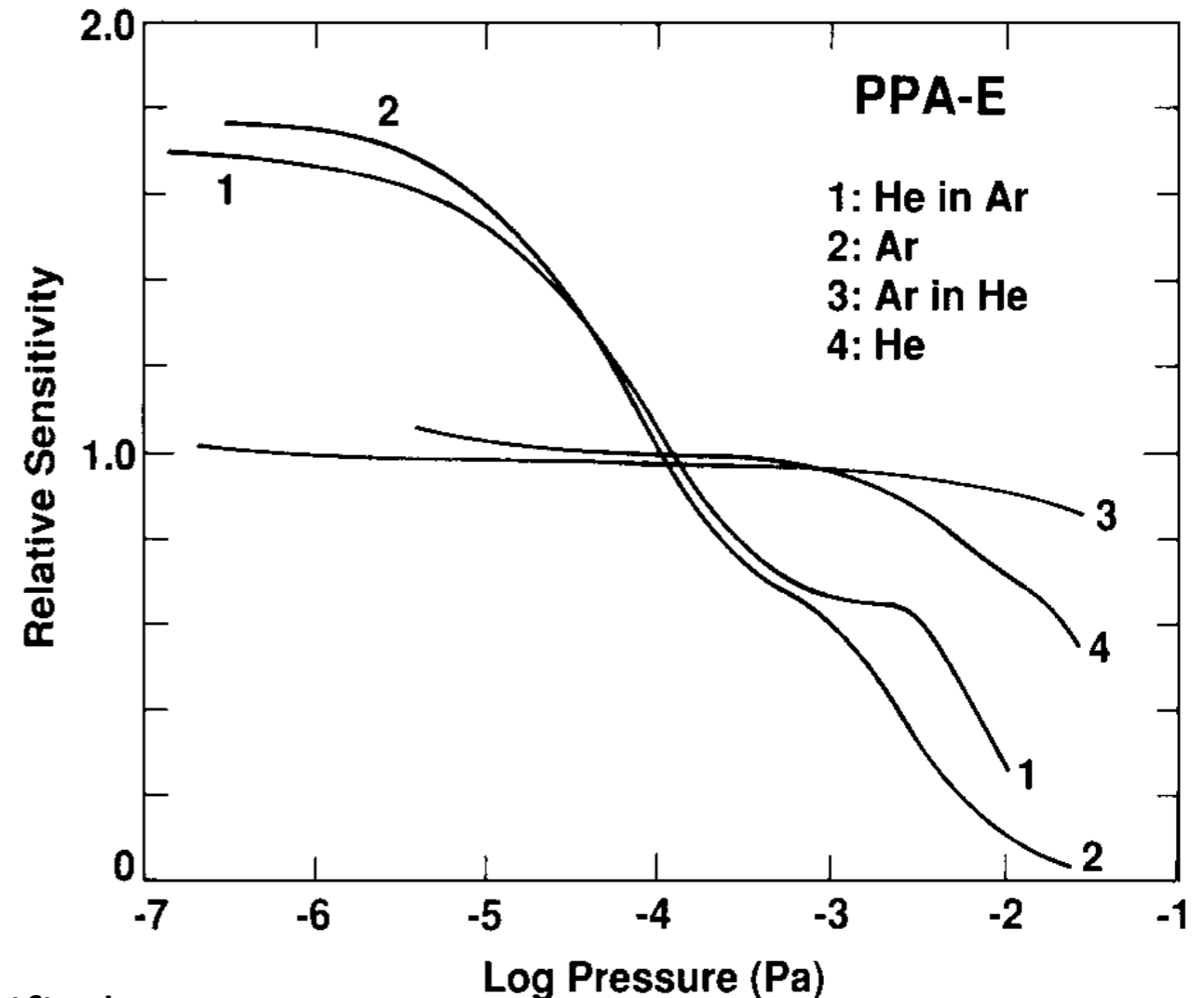
What we know

- Sensitivity depends on total pressure
- Sensitivity depends on gas mixture
- Ion source fragmentates molecules
- Ion source generates multiple charged atoms and molecules
- QMS produces and measures own mass peaks (CO, CO₂, H₂O, m/e=19)



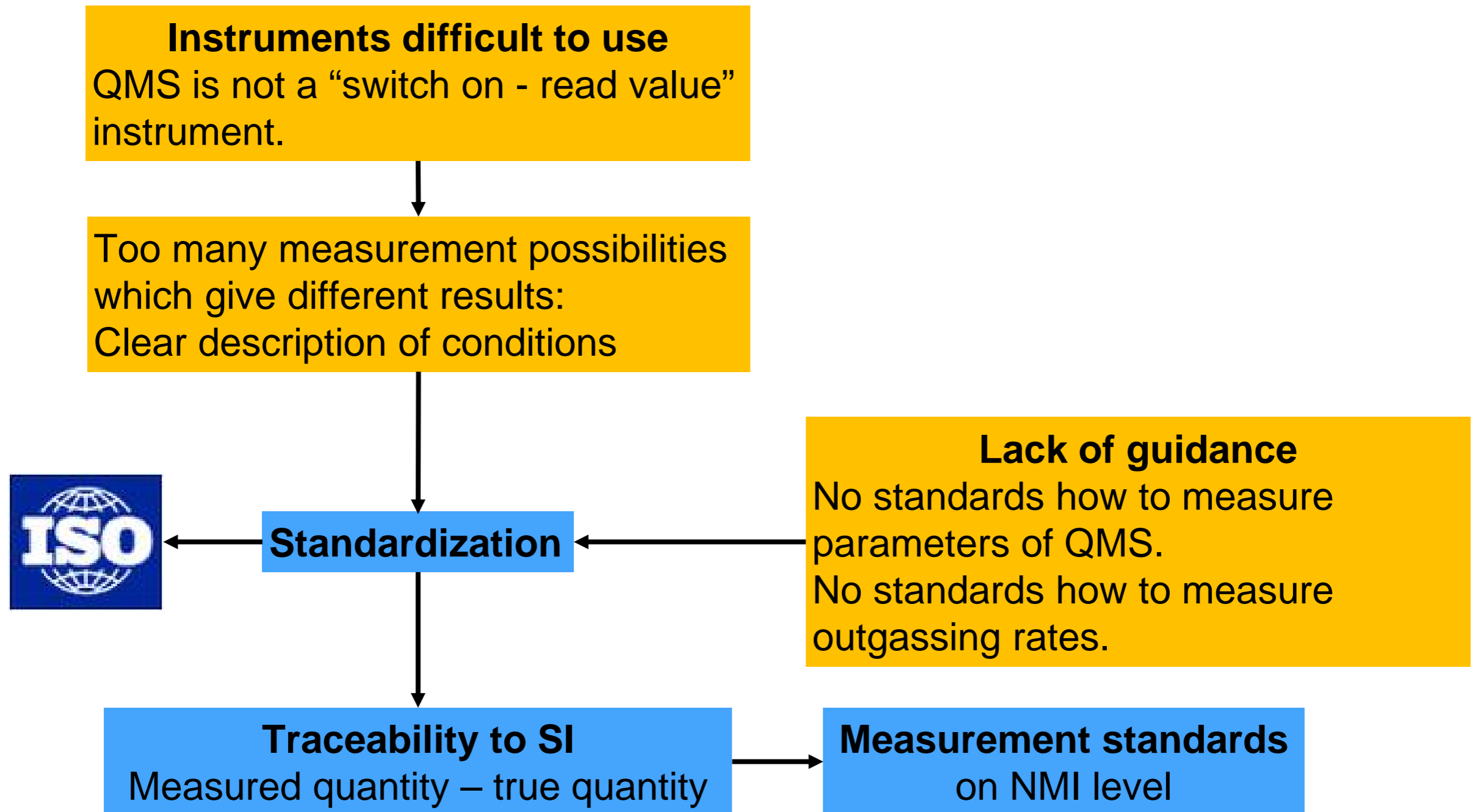
Conclusions:

- No general calibration possible
30 gas species: 1E9 possibilities to mix!
- Conditions and gases must be carefully specified
- Performance depends on application
- In-situ calibration with ion gauge helps



Lieszkovsky, Filippelli, Tilford, JVST A 8 (1990), 3838...3854

Standardization and traceability



Measurement standards on NMI level

What is needed ?

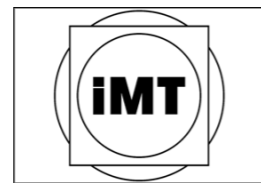
- “Fundamental” method (no simple comparison) to realize partial pressure
- Total pressure should vary between 10^{-8} Pa (ideally) to 10^{-2} Pa
- Partial pressure for at least 2 gas species, the more the better
- Partial pressure for special gas like helium $<10^{-10}$ Pa
- Known partial pressures in varying mixture (ideally 1:1 down to 10^{-6} :1)

NMIs that provide measurement standard for partial pressures
(not officially as calibration service):



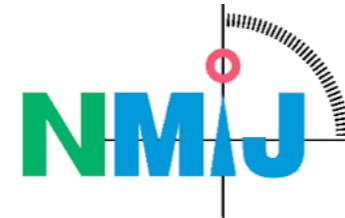
PTB, Germany

- 3 gases at the same time
- incl. H₂O and dodecane
- Total pressure 10^{-6} Pa to 10^{-2} Pa
- 1:1 down to 10^{-6} :1
- Est. 2014



IMT, Slovenia

- 3 gases at the same time
- incl. H₂O and dodecane
- 10^{-6} Pa to 10^{-2} Pa
- 1:1 down to 10^{-6} :1
- Est. 2014



NMIJ, Japan

- 2 gases at the same time
- 10^{-4} Pa to 10^{-2} Pa
- Est. 2008



NIST, USA

- 2 gases at the same time
- 10^{-6} Pa to 10^{-2} Pa
- Est. 1989 (“sleeping” status)

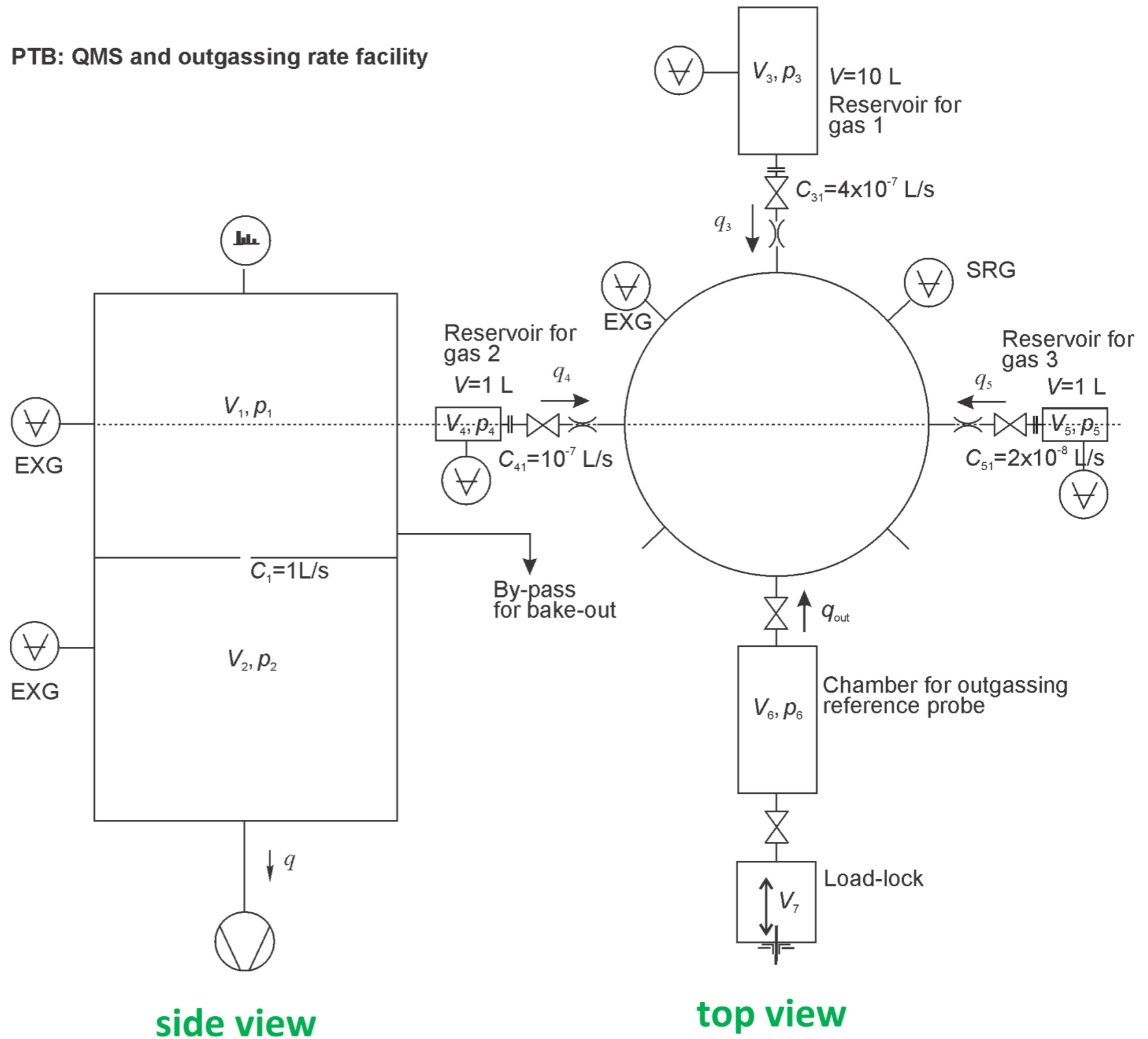
Measurement standard of PTB

PTB: QMS and outgassing rate facility

3 gases each with known partial pressure can be established in chamber V_1 at the same time

$$q_{res\ j,i} = p_{res\ j} C_{res\ j}^{molec}$$

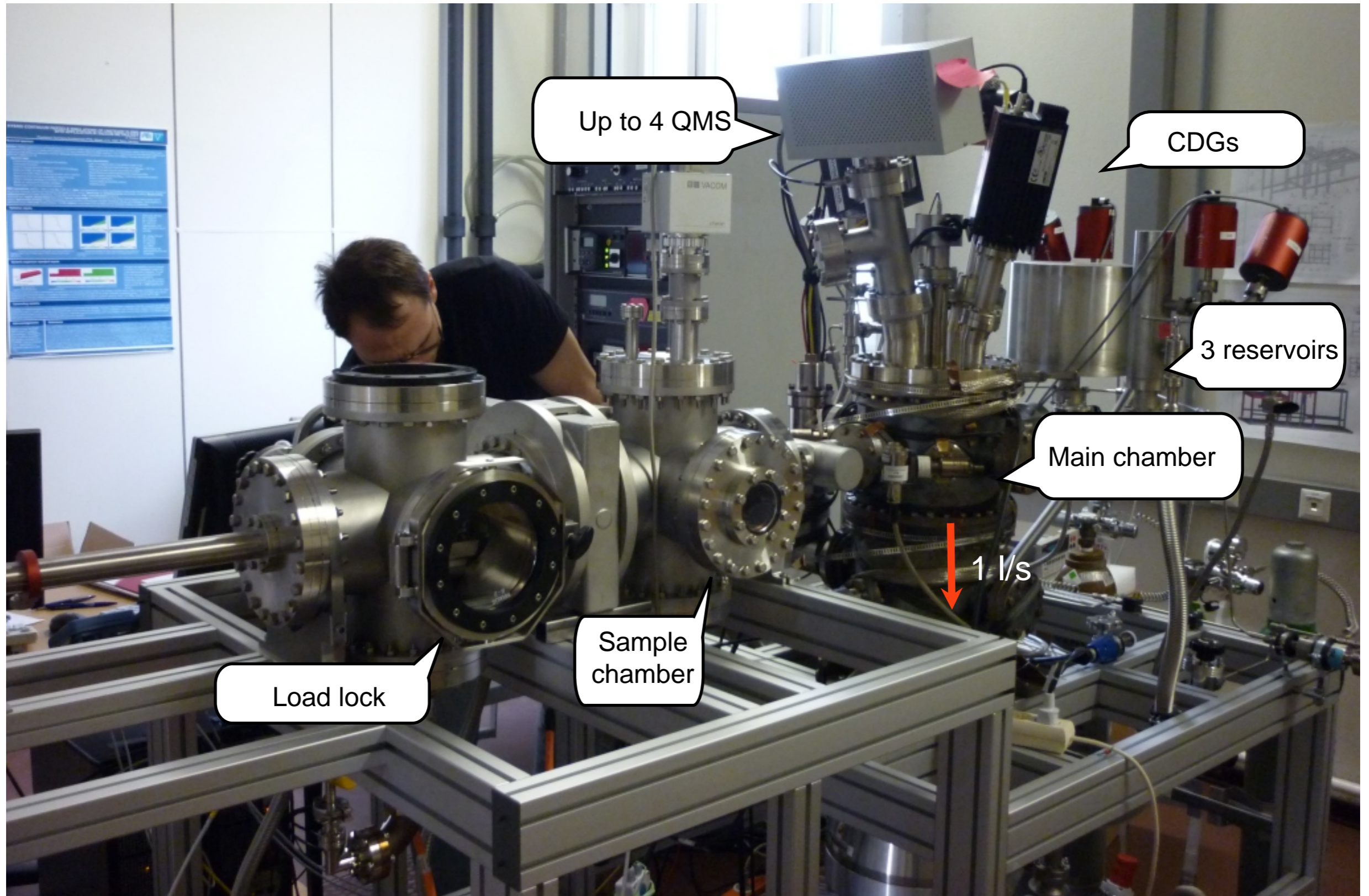
$$q_{out,i} = \frac{I_{out,i}}{I_{res\ j,i}} q_{res\ j,i}$$



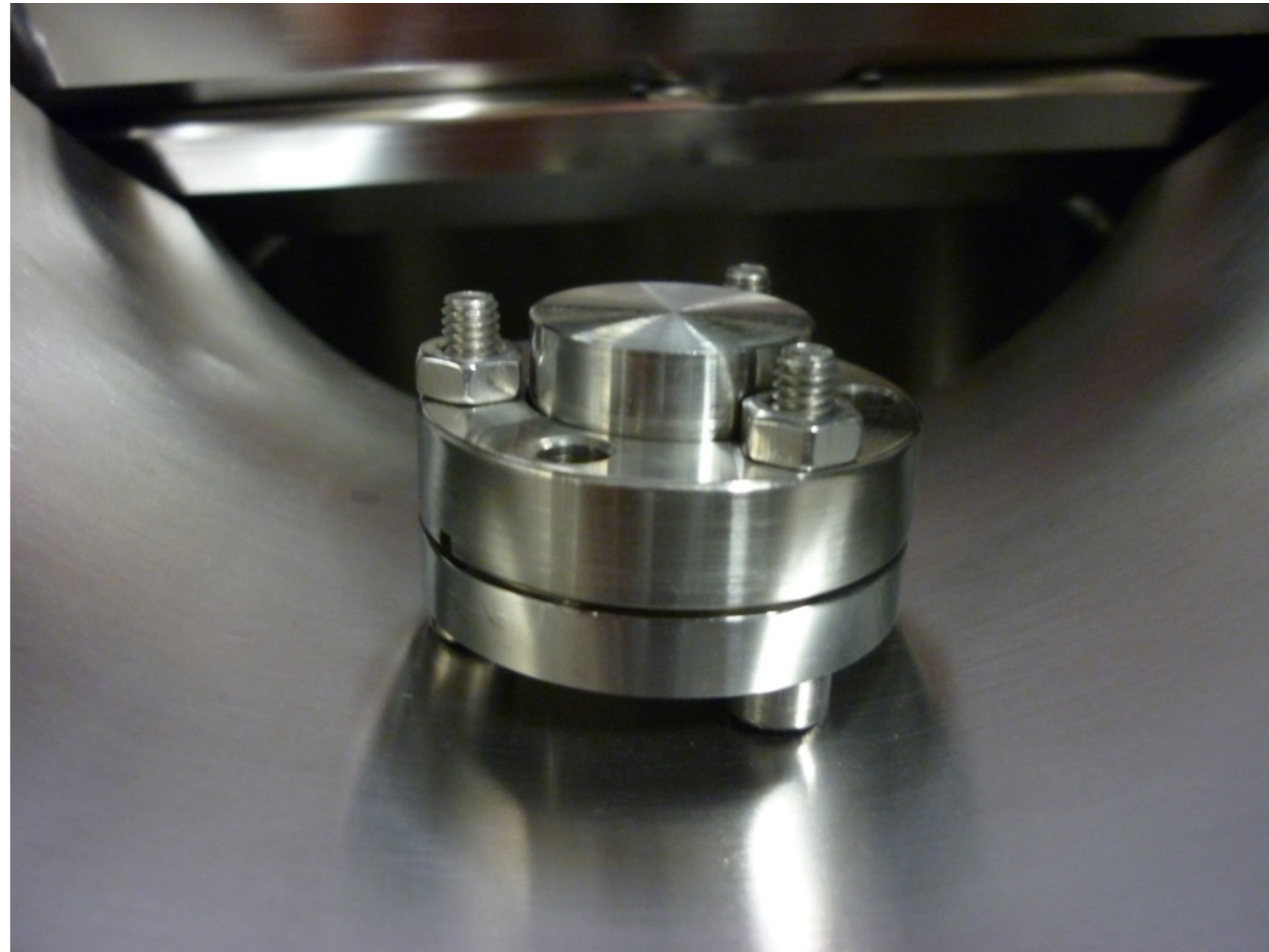
side view

top view

Measurement standard of PTB



Reference outgassing samples for comparisons of different outgassing rate measurement systems



Patent pending DE102014200907 A1 published 2015-07-23: Reference outgassing sample Setina/Jousten



TECHNICAL SPECIFICATION

ISO/TS 20175

First edition
2018-04

Vacuum technology — Vacuum gauges — Characterization of quadrupole mass spectrometers for partial pressure measurement

Technique du vide — Manomètres à vide — Description des spectromètres de masse quadripolaires pour mesurage de la pression partielle

<https://www.iso.org/standard/67207.html>

TECHNICAL SPECIFICATION

ISO/TS 20177

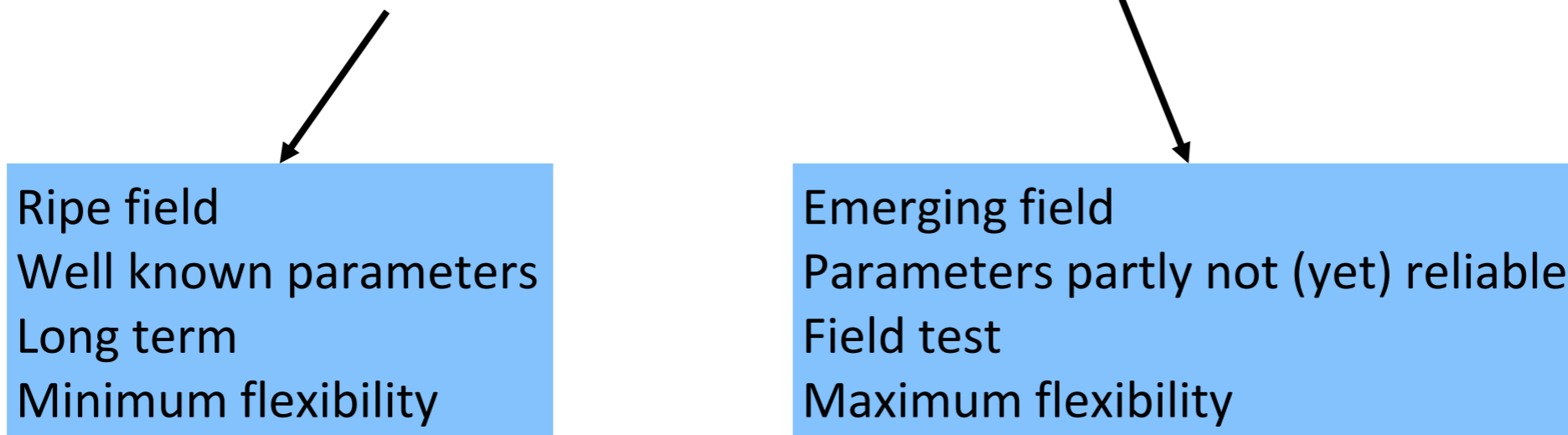
First edition
2018-06

Vacuum technology — Vacuum gauges — Procedures to measure and report outgassing rates

Technique du vide — Manomètres à vide — Méthodes de mesurage et de rapport du taux de dégagement de gaz

<https://www.iso.org/standard/67208.html>

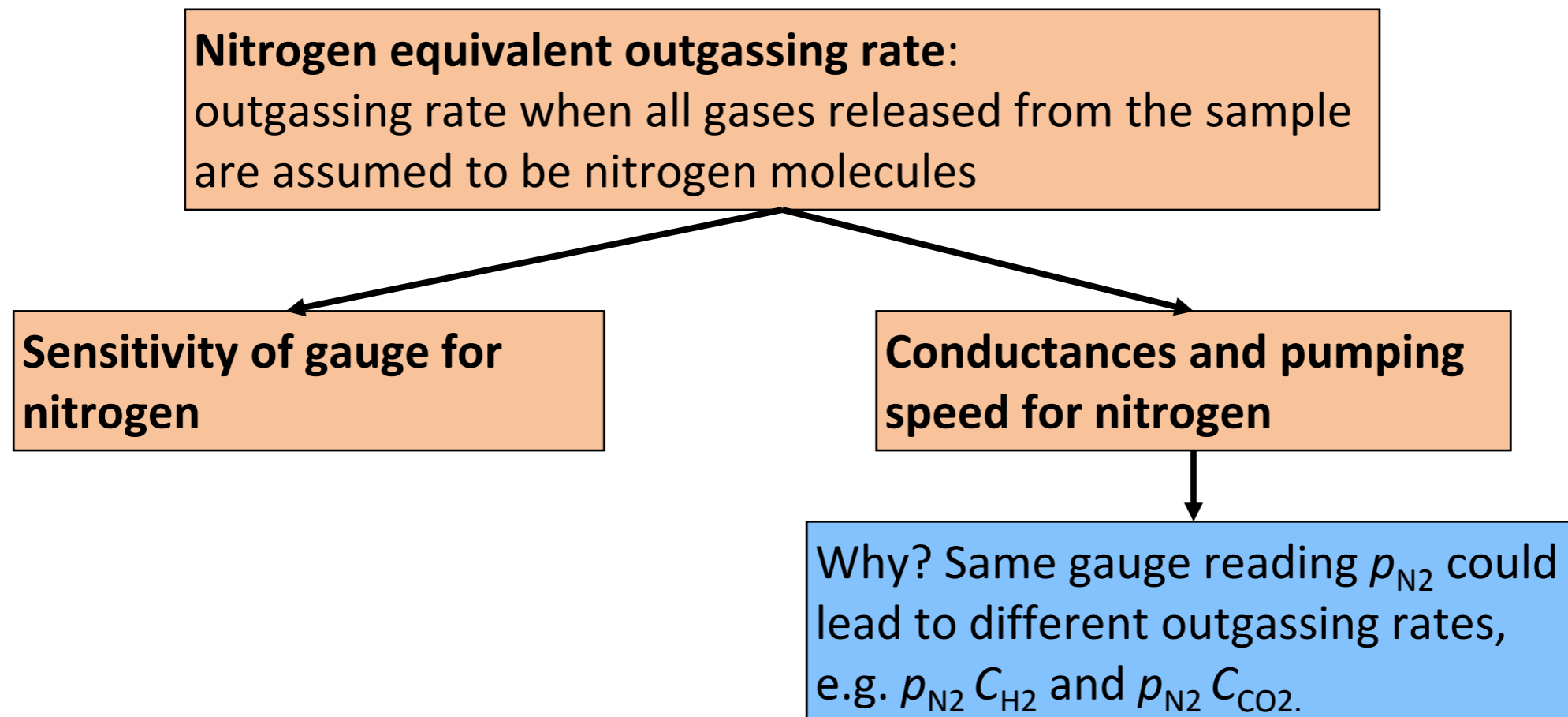
Written standard vs Technical Specification



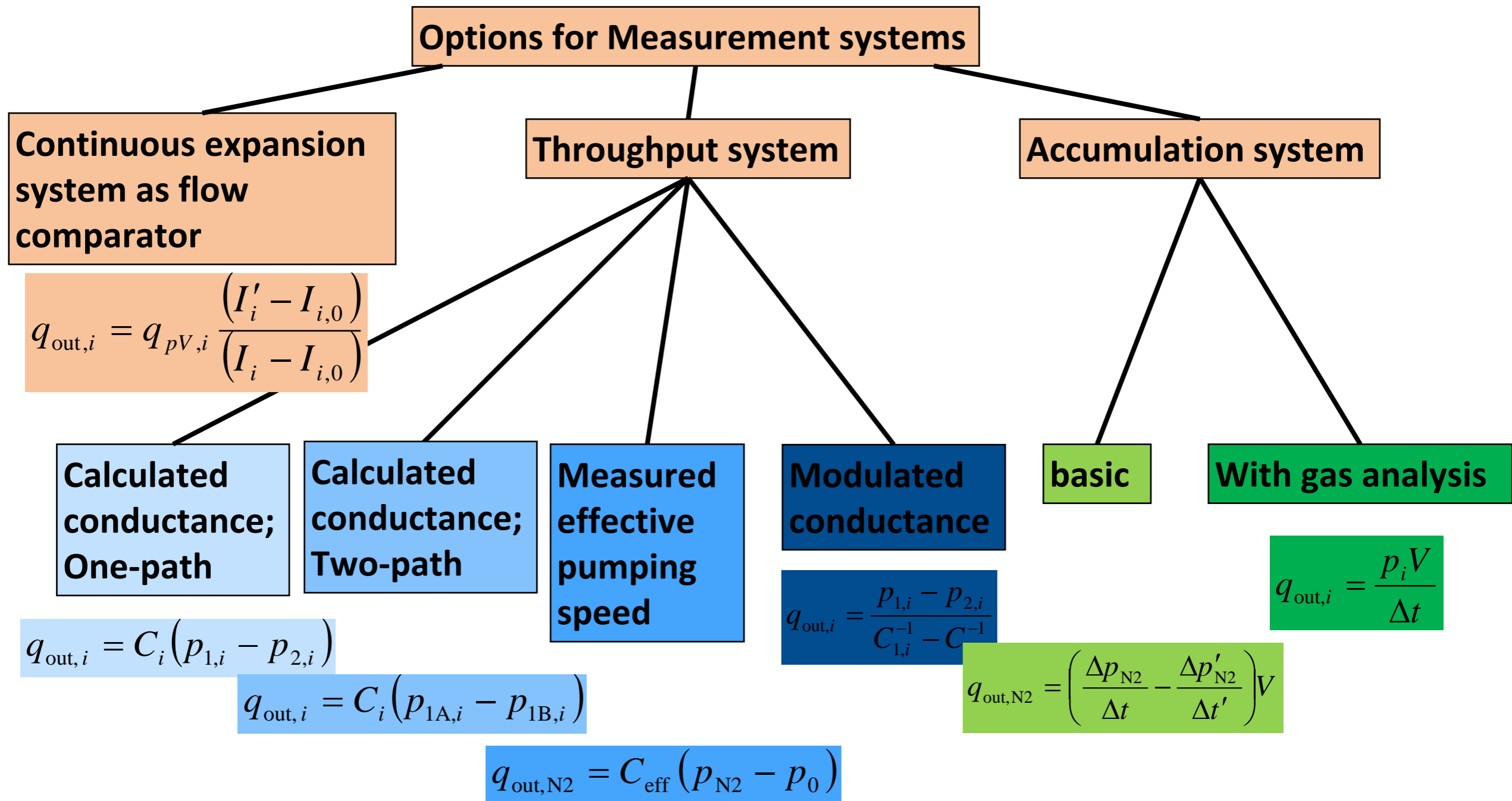
ISO Technical Specification 20177 to measure and report outgassing rates

Measurement systems

- **Several options** for measuring systems
- Some kind of **traceability** is ensured for all systems
- Concept of **nitrogen equivalent**

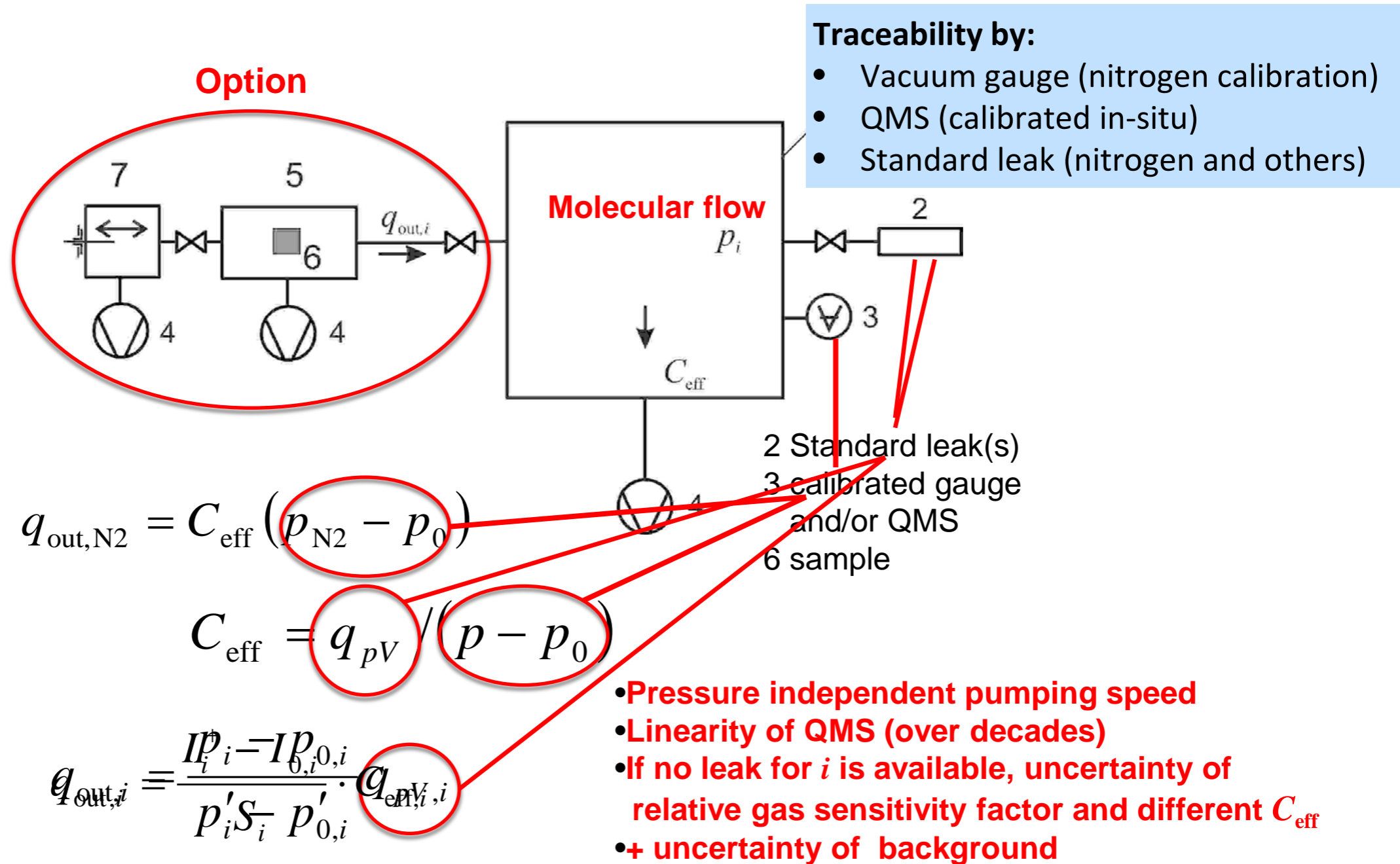


ISO Technical Specification 20177 to measure and report outgassing rates



ISO TS 20177

Throughput system with measured effective pumping speed: Explaining idea of traceability of outgassing rate



What you find in ISO TS 20177

- Which measurement system is suitable for you
- How to prepare the measurement system
- What to calibrate before
- How to prepare the sample
- How to measure
- How to report
- Measurement uncertainties

Coming up: Improve in-situ traceability of QMS

QMS have to be calibrated in-situ, not only for nitrogen

Calibrated ionization gauges are needed for many gases:
Big effort, relative gas sensitivity factors of text books inaccurate

Relative gas sensitivity factors depend on type of gauge, manufacturer, and individual gauge

Economical solution: Standardized ion gauge with no variation in relative gas sensitivity factors

European project 16NRM05
June 2017- May 2020
5 NMIs + VACOM and INFICON+
CERN+ University Lisboa

Research for standardized ion gauge

A look-up table of relative gas sensitivity factors can be produced by NMIs et al.



Support through the EMPIR 16NRM05 project is gratefully acknowledged. The EMPIR is jointly funded by the EMPIR participating countries within EURAMET and the European Union.

Conclusions

- Measurement standards for traceable desorption/ outgassing rate measurements are established
- Primary systems at NMIs exist
- Time dependent measurements can be performed
- Vacuum metrology can measure 10^{-14} kg/s
- Traceable outgassing measurements are not yet routine
- Comparison with mass loss measurements are very interesting for vacuum community



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