

# **Final report for supplementary comparison APMP.QM-S14: Hazardous air pollutants in nitrogen at 100 nmol mol<sup>-1</sup>**

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## 1. Field

Amount of substance.

## 2. Subject

Comparison of primary reference gas mixtures containing 100 nmol mol<sup>-1</sup> benzene, toluene, ethylbenzene, *m*-xylene, styrene, *o*-xylene, chloroform, 1,1,2-trichloroethane, trichloroethylene, tetrachloroethylene, 1,1-dichloroethane, carbon tetrachloride, 1,3-butadiene and dichloromethane in nitrogen.

## 3. Participants

A total of three laboratories participate in this supplementary comparison. The participants are listed in the following.

Acronym	Country	Institute
KRISS	KR	Korea Research Institute of Standards and Science, Daejeon, Republic of Korea
NIM	CN	National Institute of Metrology, Beijing, China
NMISA	ZA	National Metrology Institute of South Africa, Pretoria, South Africa

## 4. Organization body

APMP TCQM.

## 5. Introduction

Hazardous air pollutants (HAPs) are toxic air pollutants that are mainly emitted from man-made sources such as chemical processes, automobiles, power plants, and refineries, building materials, and cleaning solvents. Some HAPs are emitted from wildfires and chemical leak. HAPs are regulated and monitored to protect public health and the environment because they cause severe health effects such as cancer, damages to the immune system, reproductive and respiratory health problems.

For this supplementary comparison, a multicomponent mixture of HAPs in nitrogen has been chosen at an amount-of-substance fraction level of 100 nmol mol<sup>-1</sup> that is more close to their emission levels from their sources. The supplementary comparison is designed to underpin calibration capabilities using HAPs gas mixtures that are prepared gravimetrically as transfer standards.

## 6. Measurement schedule

Event	Deadline
Draft protocol	Apr 2016
Final protocol	Dec 2017
Registration	Jan 2018
Mixture preparation	May 2018

Shipment of mixtures	June 2018
Submission of results	Feb 2020 (NIM), Sept 2020 (NMISA)
Return of mixtures	Dec 2019
Draft A report	Oct 2020
Draft B report	Nov 2020
Final report	April 2021

## 7. Measurement standards

A suite of primary reference gas mixtures (PRMs) was prepared gravimetrically by KRISS according to ISO 6142-1[1]. Liquid raw materials for HAPs were analyzed for their purity. The diluent gases, nitrogen, were checked for their impurities. The prepared mixtures were compared against a reference gas mixture (at about 100 nmol mol<sup>-1</sup> in nitrogen) for verification. All gas mixtures were prepared in aluminum cylinders (Air Products) with Experis treatment and an internal volume of 10 dm<sup>3</sup>. The final filling pressure was approximately 10 MPa. The amount-of-substance fractions were determined based on gravimetry and purity analysis. The nominal amount-of-substance fractions were about 100 nmol mol<sup>-1</sup>.

To assign an amount-of-substance fraction to a gravimetrically prepared gas mixture, the following three groups of uncertainty components have been considered:

1. gravimetric preparation (weighing process) ( $x_{i,grav}$ )
2. purity of the parent gases ( $x_{i,purity}$ )
3. stability of the gas mixture ( $x_{i,stab}$ )

The amount-of-substance fraction,  $x_{i,prep}$ , of a target component  $i$  in mixture, can be expressed as

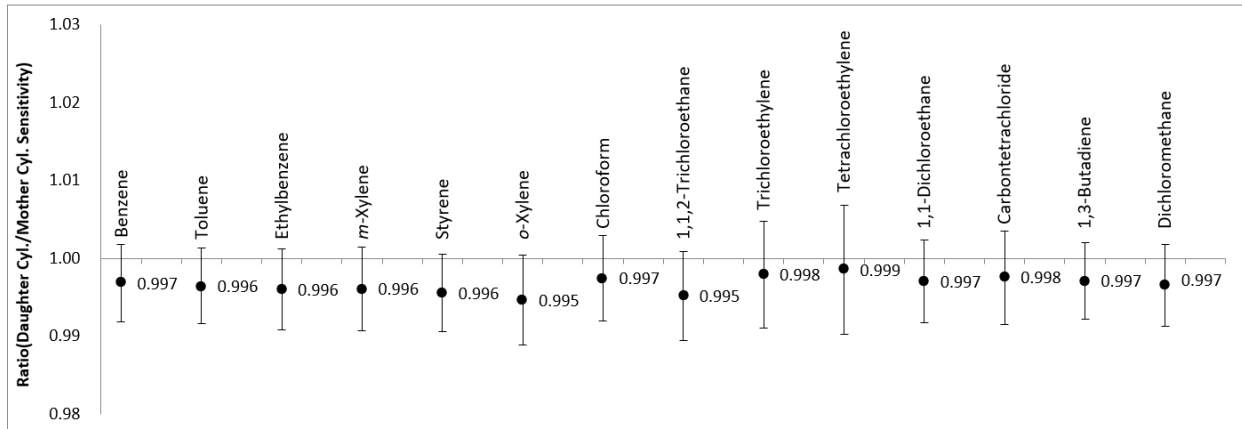
$$x_{i,prep} = x_{i,grav} + \Delta x_{i,purity} + \Delta x_{i,stab} \quad (1)$$

where  $x_{i,grav}$  is the amount-of-substance fraction of a target component  $i$  in mixture gravimetrically prepared,  $\Delta x_{i,purity}$  is the correction based on purity analysis, and  $\Delta x_{i,stab}$  is the correction due to stability. The uncertainty of the amount-of-substance fraction,  $u_{i,prep}$ , can be estimated as

$$u_{i,prep}^2 = u_{i,grav}^2 + u_{i,purity}^2 + u_{i,stab}^2 \quad (2)$$

where  $u_{i,grav}$  is the uncertainty from weighing process,  $u_{i,purity}$  is the uncertainty from purity analysis, and  $u_{i,stab}$  is the uncertainty due to short-term stability (i.e., due to physical adsorption loss onto cylinder internal surface). Physical adsorption loss for each analyte has been evaluated by applying cylinder-to-cylinder division method [2] for one of prepared gas mixtures. Results from the division method show that response ratios of mother to daughter cylinder for each analyte are not different from one statistically, indicating little physical adsorption loss for all analytes (Figure 1). The results describe that analytically determined values agree with their gravimetric values within the analytical uncertainties. Therefore, the

prepared values were not corrected due to physical adsorption loss (i.e.,  $\Delta x_{i,stab} = 0$ ). However, the differences have been used to estimate  $u_{i,stab}$  and then combined into  $u_{i,prep}$ .



**Figure 1.** Results of evaluation of physical adsorption loss onto cylinder internal surface. Error bars represent expanded uncertainties ( $k = 2$ )

The gravimetrically prepared mixtures have been verified by comparing the gravimetric value with its analytical measurement value as shown in the following conditions.

$$|x_{i,prep} - x_{i,ver}| \leq 2 \sqrt{u_{i,prep}^2 + u_{i,ver}^2} \quad (3)$$

where  $x_{i,ver}$  and  $u_{i,ver}$  are the measurement result from verification and its standard uncertainty, respectively.

The verification experiments demonstrated that the verification values agreed with the preparation values within the preparation uncertainties (Figure 2). As far as the verification experiments have demonstrated that the gravimetric values of the supplementary comparison mixtures agreed with analytical values within the uncertainty of these measurements, the reference value ( $x_{i,ref}$ ) becomes the preparation value. As a result, the standard uncertainty of the reference value is expressed as

$$u_{i,ref}^2 = u_{i,prep}^2 + u_{i,ver}^2 \quad (4)$$

To verify the sample gas mixtures prepared by KRIS throughout this comparison, each sample gas mixture was analyzed against the reference gas mixture, which was prepared with the sample gas mixtures, before their dispatch and after their return. Results from the verification showed that all gas mixtures were consistent within their measurement uncertainties (Figure 3). Although the gas mixtures agree within their uncertainties, an excess uncertainty for long-term stability was added into the verification uncertainties (Table 2). The maximum difference in the verification results for each component was added as the excess uncertainty. The excess standard uncertainty was estimated by dividing the maximum difference by  $2\sqrt{3}$  assuming rectangular distribution.

All gas mixtures at  $100 \text{ nmol mol}^{-1}$  were analyzed using a GC-FID with a cryogenic pre-concentrator. The detailed analytical conditions are summarized in the following.

GC-FID

Column	VOCOL (60 m × 320 μm × 1.8 μm)
	Flow 2 mL/min (He)
Oven	105 °C (isothermal, 34.5 min)
Detector	250 °C (isothermal), H <sub>2</sub> : 35 mL/min, Air: 300 mL/min

	Trap temperature, °C	M1→M2 temperature, °C	M2→M3 temperature, °C	Inject temperature, °C	Bake out temperature, °C
Mod 1 Trap	40	40			40
Mod 2 Trap	-95	-95	220		200
Mod 3 Trap			-170	110	
Trapping sample					
Sample flow	90 mL/min				
Sample volume	270 mL				

## 8. Measurement protocols

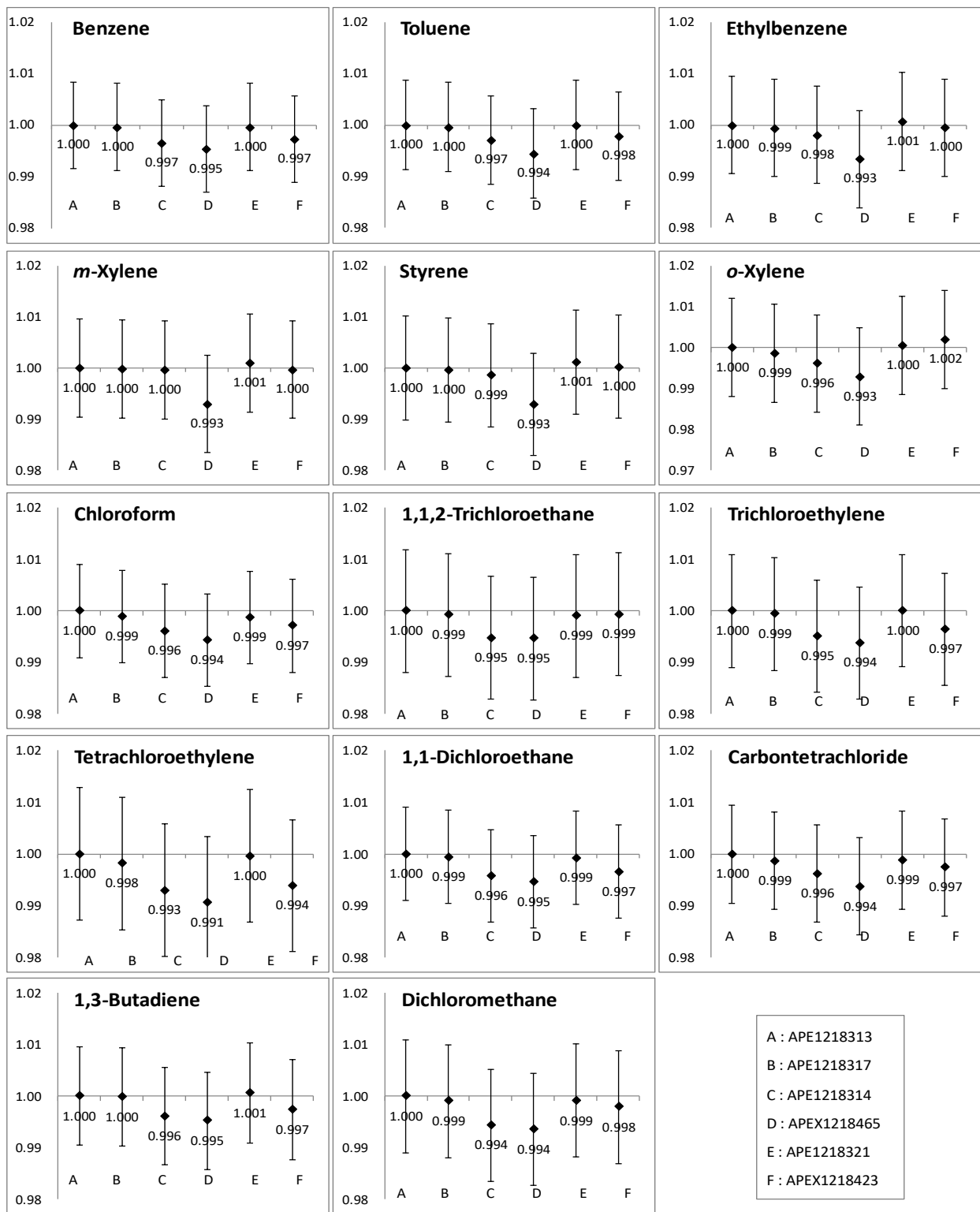
Each laboratory was requested to perform at least 3 measurements with independent calibrations. All laboratories were also asked to provide detailed information regarding their calibration standards, analytical method, and uncertainty evaluation.

## 9. Measurement method

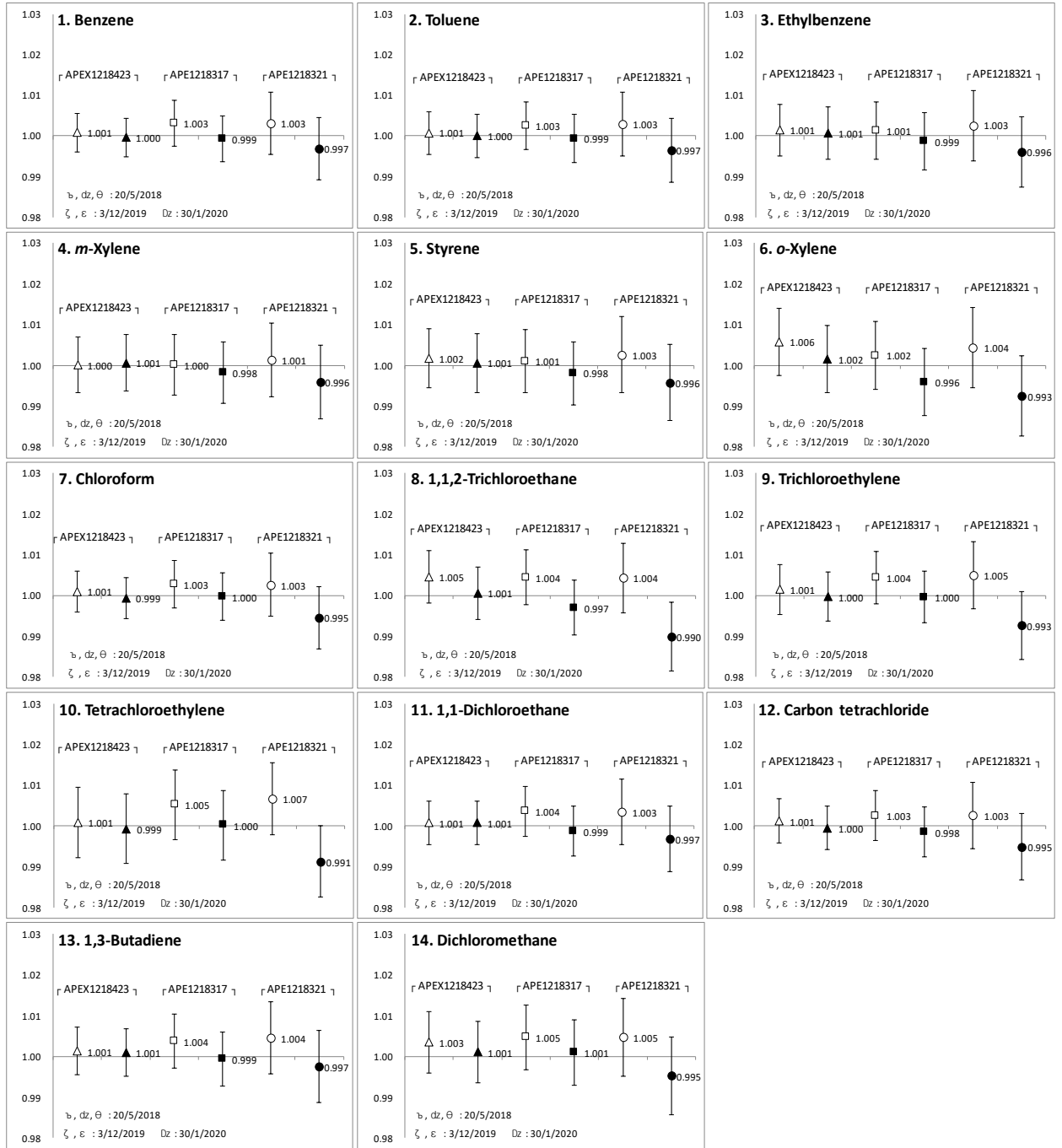
The details on the measurement methods used by the participants are described in the individual participant reports. A summary of the calibration method, date of measurement, and the way in which metrological traceability is established is given in Table 1.

**Table 1.** Summary of the measurement methods of the participants

Laboratory	Cylinder	Measurement period	Calibration standards	Instrument calibration	Measurement technique
KRISS	APEX1218423	Aug 2018	Own standards	one point calibration	GC-FID with pre-concentrator
NIM	APE1218317	Apr 2018	Own standards	one point calibration	GC-MSD
NMISA	APE1218321	May 2018	Own standards	one point calibration	GC-FID with pre-concentrator



**Figure 2.** Results from verification of gravimetrically prepared gas mixtures. Error bars represent expanded uncertainties ( $k = 2$ ) and y-axis is for the sensitivity ratio of sample to reference gas mixture



**Figure 3.** Results of verification before and after dispatching sample cylinders. Error bars represent expanded uncertainties ( $k = 2$ ) and y-axis is for the sensitivity ratio of sample to reference gas mixture.

## 10. Degree of equivalence (DoE)

A degree of equivalence for each participating laboratory was calculated as

$$D_i = x_{i,lab} - x_{i,SCRV} \quad (5)$$

where  $x_{i,lab}$  and  $x_{i,SCRV}$  are the value reported by each participant and the supplementary comparison reference value (SCRV), respectively. In this comparison, the preparation value is set to the SCRV value as expressed in the following.



$$x_{i,SCRV} = x_{i,ref} \quad (6)$$

Thus, the uncertainty of the SCRV values can be expressed as

$$u_{i,SCRV} = u_{i,ref} \quad (7)$$

Therefore, the standard uncertainty of  $D_i$  can be expressed as

$$u^2(D_i) = u_{i,lab}^2 + u_{i,SCRV}^2 \quad (8)$$

where  $u_{i,lab}$  and  $u_{i,SCRV}$  are the standard uncertainties of  $x_{i,lab}$  and  $x_{i,SCRV}$ , respectively

## 11. Results and Discussion

A complete set of results from each participant is described in annex A of this report. The results of the supplementary comparison are summarized in Table 2 and Figure 4.

**Table 2.** Summarized results for APMP.QM-S14 (nmol mol<sup>-1</sup>)

Laboratory Cylinder	$x_{i,ref}$	$u_{i,prep}$	$u_{i,ver}$	$u_{i,ref}$	$x_{i,lab}$	$u_{i,lab}$	$D_i$	$U(D_i)$ $k = 2$
<b>NIM –</b>								
<b>APE1218317</b>								
benzene	104.20	0.11	0.43	0.44	105.2	1.1	1.0	2.3
toluene	104.73	0.12	0.44	0.45	105.2	1.1	0.5	2.3
ethylbenzene	103.94	0.15	0.47	0.49	104.3	1.0	0.4	2.3
m-xylene	103.10	0.16	0.47	0.49	103.3	1.5	0.2	3.3
styrene	102.73	0.18	0.49	0.52	102.5	1.5	-0.2	3.2
o-xylene	103.59	0.19	0.59	0.62	102.0	1.5	-1.6	3.3
chloroform	106.11	0.13	0.46	0.48	108.6	1.1	2.5	2.4
1,1,2-trichloroethane	101.56	0.17	0.58	0.61	101.7	1.5	0.1	3.3
trichloroethylene	104.13	0.07	0.56	0.57	105.3	1.6	1.2	3.4
tetrachloroethylene	101.53	0.07	0.65	0.65	101.9	1.0	0.4	2.4
1,1-dichloroethane	103.54	0.13	0.45	0.47	103.9	1.0	0.4	2.3
carbon tetrachloride	102.99	0.16	0.46	0.49	102.2	1.0	-0.8	2.3
1,3-butadiene	102.86	0.17	0.46	0.49	102.8	1.0	-0.1	2.3
dichloromethane	102.69	0.24	0.51	0.51	102.1	1.0	-0.6	2.3
<b>NMISA –</b>								
<b>APE1218321</b>								
benzene	105.42	0.11	0.43	0.44	118.6	3.5	13.2	7.0
toluene	105.96	0.12	0.44	0.46	108.3	3.4	2.3	6.9
ethylbenzene	105.16	0.15	0.47	0.50	112.6	3.0	7.4	6.0
m-xylene	104.31	0.16	0.47	0.50	116.4	2.8	12.1	5.6

styrene	103.93	0.18	0.49	0.53	*	*		
o-xylene	104.81	0.19	0.60	0.63	93.8	2.5	-11.0	5.1
chloroform	106.78	0.13	0.46	0.48	109.1	2.2	2.3	4.4
1,1,2-trichloroethane	102.20	0.18	0.58	0.61	*	*		
trichloroethylene	104.79	0.07	0.57	0.57	103.3	2.9	-1.5	5.8
tetrachloroethylene	102.17	0.07	0.65	0.66	128.0	6.0	25.8	12.1
1,1-dichloroethane	104.20	0.13	0.45	0.47	*	*		
carbon tetrachloride	103.64	0.16	0.46	0.49	*	*		
1,3-butadiene	104.27	0.19	0.47	0.50	106.1	0.7	1.8	1.6
dichloromethane	107.85	0.25	0.54	0.59	103.2	2.9	-4.6	5.9

**KRISS –  
APEX1218423**

benzene	104.68	0.11	0.43	0.44	104.61	0.47	-0.07	1.29
toluene	105.21	0.12	0.44	0.46	105.20	0.49	-0.01	1.33
ethylbenzene	104.42	0.15	0.47	0.49	104.53	0.52	0.11	1.44
m-xylene	103.58	0.16	0.47	0.50	103.66	0.55	0.08	1.47
styrene	103.20	0.18	0.49	0.52	103.30	0.57	0.10	1.55
o-xylene	104.07	0.19	0.59	0.62	104.39	0.69	0.32	1.86
chloroform	106.03	0.13	0.46	0.48	105.93	0.51	-0.10	1.39
1,1,2-trichloroethane	101.48	0.17	0.58	0.61	101.56	0.64	0.07	1.76
trichloroethylene	104.06	0.07	0.56	0.57	103.99	0.59	-0.07	1.64
tetrachloroethylene	101.45	0.07	0.65	0.63	101.41	0.67	-0.05	1.88
1,1-dichloroethane	103.46	0.13	0.45	0.47	103.51	0.51	0.05	1.38
carbon tetrachloride	102.91	0.16	0.46	0.49	102.80	0.54	-0.11	1.46
1,3-butadiene	103.53	0.19	0.46	0.50	103.58	0.51	0.04	1.44
dichloromethane	107.09	0.25	0.53	0.59	107.17	0.61	0.08	1.69

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\* Note: the analytes were not measured by NMISA

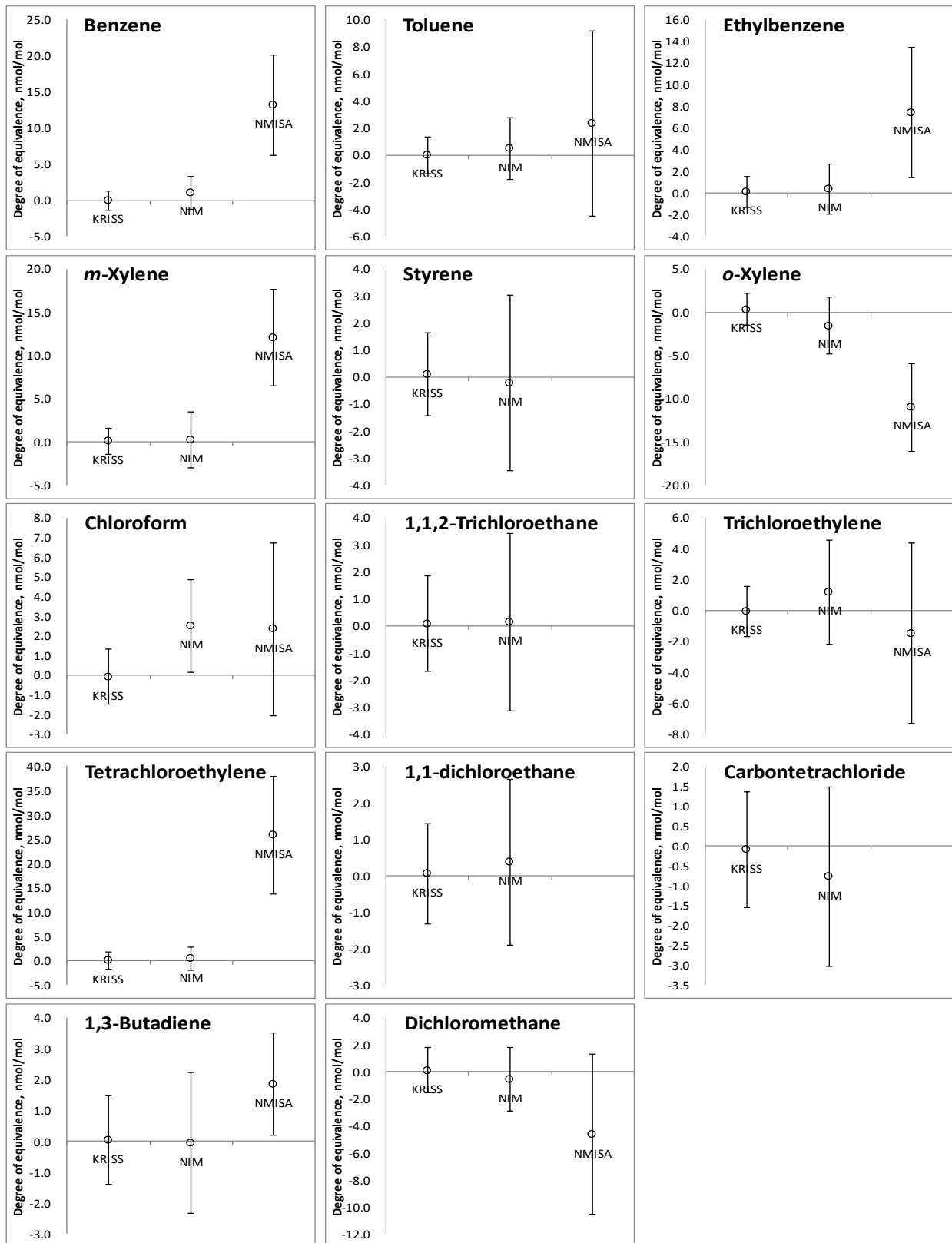


Figure 4. Degrees of equivalence for APMP.QM-S14. Error bars represent expanded uncertainties ( $k = 2$ )

## 12. Conclusions

The comparison has been successfully conducted to evaluate the participants' measurement capabilities for HAPs in nitrogen at  $100 \text{ nmol mol}^{-1}$ . Both NIM and KRISS results are consistent with their SCRVs for all analytes except chloroform from NIM. NMISA results are consistent with their SCRVs for toluene, styrene, chloroform, 1,1,2-trichloroethane, trichloroethylene, 1,1-dichloroethane, carbon tetrachloride, and dichloromethane while NMISA results are not consistent with their SCRVs for benzene, ethylbenzene, m-xylene, o-xylene, tetrachloroethylene, 1,3-butadiene.

## 13. Supported CMC claims

This supplementary comparison underpins core skills and competencies required in gravimetric preparation, analytical verification and purity analysis of HAPs. It is considered as a Track C comparison due to its nature with the stability challenges. The results of this supplementary comparison can be used to support CMC claims for benzene, toluene, ethylbenzene, m-xylene, styrene, o-xylene, chloroform, 1,1,2-trichloroethane, trichloroethylene, tetrachloroethylene, 1,1-dichloroethane, 1,2-dichloroethane, carbon tetrachloride, 1,3-butadiene and dichloromethane in nitrogen as explained in the Annex A.

## 14. References

- [1] ISO 6142-1 2015, *Gas analysis – Preparation of calibration gas mixtures – Part 1: Gravimetric method for Class I mixtures*.
- [2] Lee S, Kim M E, Oh S H, Kim J S 2017 Determination of physical adsorption loss of primary standard gas mixtures in cylinders using cylinder-to-cylinder division *Metrologia* **54** L26.

### Coordinator and Correspondence

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Gas Metrology Group

Korea Research Institute of Standards and Science

## Annex A: Examples for HFTLS statements for CMC claims

### 1. Participant's results are consistent with SCRVs

When a participant reported standard uncertainty is  $0.4 \text{ nmol mol}^{-1}$  and its SCRv is  $106.34 \text{ nmol mol}^{-1}$ , its HFTLS statements are in the following.

Lower limit of amount fraction ( $\text{nmol mol}^{-1}$ )	Upper limit of amount fraction ( $\text{nmol mol}^{-1}$ )	Lower limit of expanded uncertainty (%)	Upper limit of expanded uncertainty (%)
0.8	100	100	0.8
100	10 000	0.8	0.8

### 2. Participant's results are not consistent with SCRVs

When a participant reported standard uncertainty is  $1.1 \text{ nmol mol}^{-1}$ , its SCRv is  $104.8 \text{ nmol mol}^{-1}$ , and D is  $6.3 \text{ nmol mol}^{-1}$ , its HFTLS statements are in the following.

Lower limit of amount fraction ( $\text{nmol mol}^{-1}$ )	Upper limit of amount fraction ( $\text{nmol mol}^{-1}$ )	Lower limit of expanded uncertainty (%)	Upper limit of expanded uncertainty (%)
12.8	100	100	12.2
100	10 000	12.2	12.2

## Annex B: Measurement reports

### Report form

Laboratory name: National Institute of Metrology, China

Cylinder number: APE1218317

#### Measurement 1<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol mol <sup>-1</sup> )	Standard deviation (% relative)	Number of replicates
Benzene	15/08/2019	104.6	0.10%	6
Toluene	15/08/2019	104.6	0.17%	6
Ethylbenzene	15/08/2019	103.6	0.12%	6
o-xylene	15/08/2019	101.5	0.18%	6
m-xylene	15/08/2019	102.8	0.22%	6
Styrene	15/08/2019	100.1	0.24%	6
1,3-butadiene	15/08/2019	102.6	0.15%	6
Tetrachloroethylene	15/08/2019	101.4	0.30%	6
Trichloroethylene	15/08/2019	106.1	0.23%	6
1,1,2-trichloroethane	15/08/2019	100.9	0.21%	6
1,1-dichloroethane	15/08/2019	103.4	0.29%	6
Carbon tetrachloride	15/08/2019	102.2	0.32%	6
Chloroform	15/08/2019	108.6	0.34%	6
Dichloromethane	15/08/2019	102.0	0.34%	6

#### Measurement 2<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol mol <sup>-1</sup> )	Standard deviation (% relative)	Number of replicates
Benzene	21/08/2019	105.5	0.36%	6
Toluene	21/08/2019	105.3	0.28%	6
Ethylbenzene	21/08/2019	103.8	0.47%	6
o-xylene	21/08/2019	100.5	0.29%	6
m-xylene	21/08/2019	104.0	0.35%	6
Styrene	21/08/2019	103.8	0.26%	6
1,3-butadiene	21/08/2019	103.0	0.43%	6
Tetrachloroethylene	21/08/2019	101.8	0.33%	6
Trichloroethylene	21/08/2019	104.2	0.23%	6
1,1,2-trichloroethane	21/08/2019	101.6	0.26%	6
1,1-dichloroethane	21/08/2019	103.9	0.29%	6
Carbon tetrachloride	21/08/2019	101.8	0.25%	6
Chloroform	21/08/2019	108.1	0.20%	6
Dichloromethane	21/08/2019	102.0	0.39%	6

#### Measurement 3<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol mol <sup>-1</sup> )	Standard deviation (% relative)	Number of replicates
Benzene	25/08/2019	106.0	0.28%	6
Toluene	25/08/2019	105.8	0.09%	6
Ethylbenzene	25/08/2019	104.8	0.24%	6
o-xylene	25/08/2019	103.2	0.08%	6
m-xylene	25/08/2019	103.6	0.23%	6
Styrene	25/08/2019	103.5	0.26%	6
1,3-butadiene	25/08/2019	102.1	0.06%	6
Tetrachloroethylene	25/08/2019	102.2	0.16%	6
Trichloroethylene	25/08/2019	104.4	0.15%	6
1,1,2-trichloroethane	25/08/2019	102.2	0.19%	6
1,1-dichloroethane	25/08/2019	104.2	0.14%	6
Carbon tetrachloride	25/08/2019	101.8	0.19%	6
Chloroform	25/08/2019	108.8	0.12%	6
Dichloromethane	25/08/2019	102.1	0.25%	6

#### Measurement 4

Component	Date (dd/mm/yy)	Result (nmol mol <sup>-1</sup> )	Standard deviation (% relative)	Number of replicates
Benzene	16/09/2019	104.7	0.23%	6
Toluene	16/09/2019	105.1	0.30%	6
Ethylbenzene	16/09/2019	104.8	0.31%	6
o-xylene	16/09/2019	102.7	0.27%	6
m-xylene	16/09/2019	102.9	0.16%	6
Styrene	16/09/2019	102.4	0.80%	6
1,3-butadiene	16/09/2019	103.6	0.47%	6
Tetrachloroethylene	16/09/2019	102.1	0.56%	6
Trichloroethylene	16/09/2019	106.3	0.41%	6
1,1,2-trichloroethane	16/09/2019	102.0	0.39%	6
1,1-dichloroethane	16/09/2019	103.9	0.53%	6
Carbon tetrachloride	16/09/2019	102.9	0.59%	6
Chloroform	16/09/2019	108.7	0.53%	6
Dichloromethane	16/09/2019	102.4	0.53%	6

#### Results

Component	Result (nmol mol <sup>-1</sup> )	Expanded uncertainty (nmol mol <sup>-1</sup> )	Coverage factor
Benzene	105.2	2%	2
Toluene	105.2	2%	2
Ethylbenzene	104.3	2%	2
o-xylene	102.0	3%	2

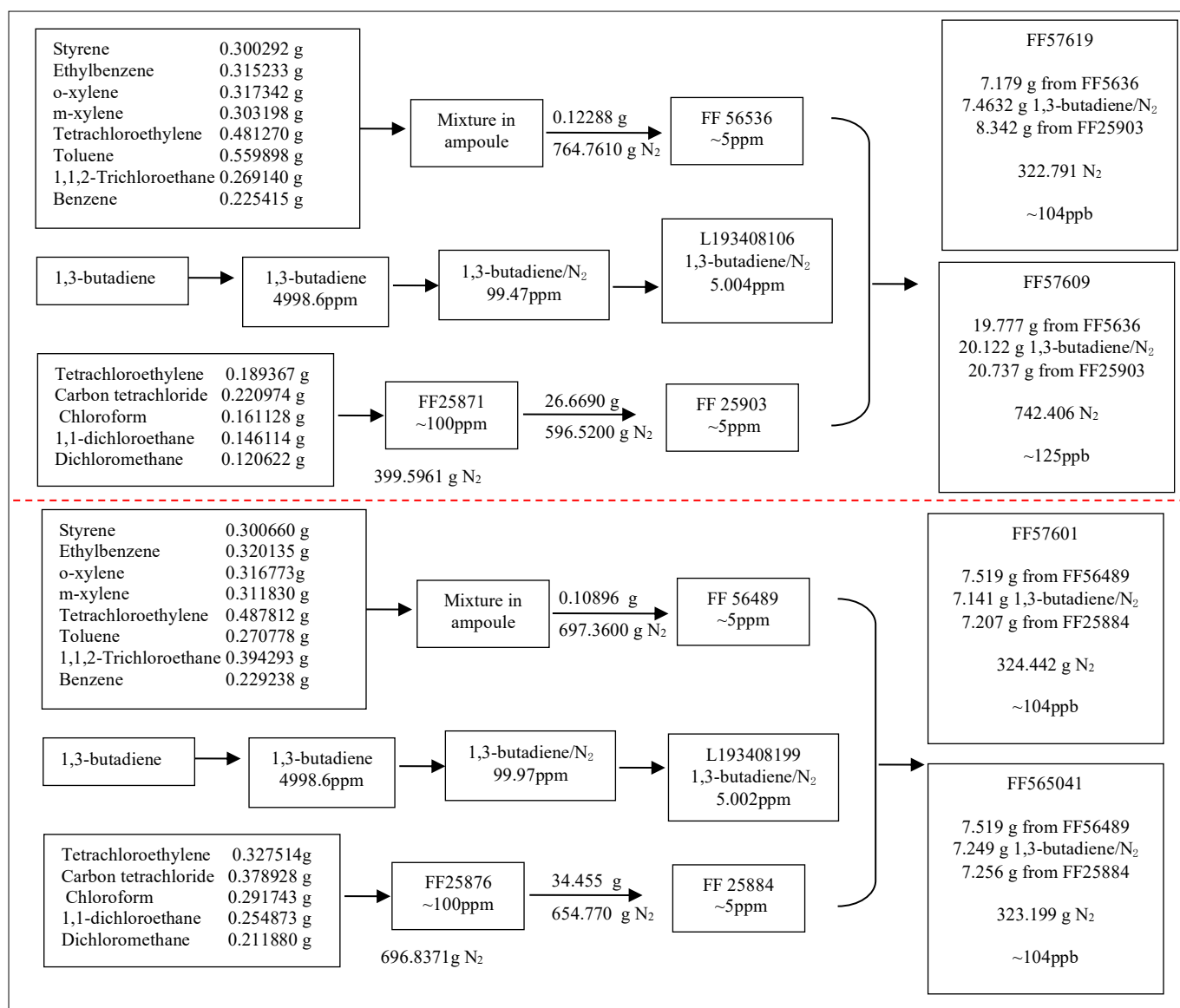
m-xylene	103.3	3%	2
Styrene	102.5	3%	2
1,3-butadiene	102.8	2%	2
Tetrachloroethylene	101.9	2%	2
Trichloroethylene	105.3	3%	2
1,1,2-trichloroethane	101.7	3%	2
1,1-dichloroethane	103.9	2%	2
Carbon tetrachloride	102.2	2%	2
Chloroform	108.6	2%	2
Dichloromethane	102.1	2%	2

### Calibration standards

Please provide a brief description of the calibration standards used.

The following chart shows a brief outline of the dilution series undertaken to produce the final mixtures. The dilution series were designed based on the saturated vapor pressure of the target components. The components with similar saturated vapor pressure were mixed together so that the evaporation loss was minimized. The components were either directly injected into the cylinder or mixed in ampoule prior to the injection. The calibration standards were measured against each other and the consistency of different cylinder was within 0.1%-0.7%.





## Analytical method

Please provide a brief description of the instrumentation and method used for analysis

The calibration standard and sample were directly injected into GCMS system without preconcentration. An A-B-A sequence were repeated several times so that the instrument reading was stabilized to an acceptable level. The chromatographic condition is listed as below:

System	Injection volume	Split ratio	Oven temperature	Column	Ion mode
GCMS	5mL	30:1	40°C 5min 5°C/min to 180 °C	DB624UI 60m*0.32mm*1.8µm	SIM

## Uncertainty evaluation

Please provide a brief description of the evaluation of measurement uncertainty.

The major uncertainty came from the following parts: the purity of materials, the weighing, the atomic weight as well as the consistency of verification. The effect of adsorption was also evaluated.

**a) Purity table for the nominally pure Benzene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00170	0.00017
Others	GCMS	Normal	0.00220	0.00022
Benzene	/	/	0.99610	0.00028

**b) Purity table for the nominally pure Toluene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00080	0.00008
Others	GCMS	Normal	0.00140	0.00014
Toluene	/	/	0.99860	0.00016

**c) Purity table for the nominally pure Ethylbenzene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Toluene	GCMS	Normal	0.00230	0.00023
Water	Karl Fischer	Normal	0.00179	0.00018
Benzene	GCMS	Normal	0.00160	0.00016
Others	GCMS	Normal	0.00530	0.00053
Ethylbenzene	/	/	0.98901	0.00063

**d) Purity table for the nominally pure o-xylene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00100	0.00010
m-xylene	GCMS	Normal	0.01260	0.00126
Others	GCMS	Normal	0.00430	0.00043
o-xylene	/	/	0.98210	0.00134

**e) Purity table for the nominally pure m-xylene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	GCMS	Normal	0.00100	0.00010
p-xylene	Karl Fischer	Normal	0.01260	0.00126
Others	GCMS	Normal	0.00430	0.00043
o-xylene	GCMS	Normal	0.98210	0.00134

**f) Purity table for the nominally pure Styrene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
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Water	Karl Fischer	Normal	0.23320	0.02332
Others	GCMS	Normal	0.05000	0.00500
Styrene	/	/	0.99717	0.00024

**g) Purity table for the nominally pure 1,3-butadiene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
N <sub>2</sub>	PDHID	Normal	0.00005	0.00001
Others	GCMS	Normal	0.00015	0.00002
1,3-butadiene	/	/	0.99980	0.00002

**h) Purity table for the nominally pure Tetrachloroethylene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00088	0.00009
Others	GCMS	Normal	0.00040	0.00004
Tetrachloroethylene	/	/	0.99872	0.00010

**i) Purity table for the nominally pure Trichloroethylene**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00106	0.00011
Others	GCMS	Normal	0.00280	0.00028
Trichloroethylene	/	/	0.99614	0.00030

**j) Purity table for the nominally pure 1,1,2-trichloroethane**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00354	0.00035
Others	GCMS	Normal	0.01240	0.00124
1,1,2-trichloroethane	/	/	0.98406	0.03545

**k) Purity table for the nominally pure 1,1-dichloroethane**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00148	0.00015
Others	GCMS	Normal	0.01790	0.00179
1,1-dichloroethane	/	/	0.98062	0.0018

**l) Purity table for the nominally pure Carbon tetrachloride**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00072	0.00007
Others	GCMS	Normal	0.00520	0.00052
CCl <sub>4</sub>	/	/	0.99408	0.00052

**m) Purity table for the nominally pure Chloroform**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00106	0.00011
Others	GCMS	Normal	0.00280	0.00028
Chloroform	/	/	0.99614	0.00030

**n) Purity table for the nominally pure Dichloromethane**

Compounds	Instrument	Distribution	Concentration mol/mol	Standard uncertainty
Water	Karl Fischer	Normal	0.00191	0.00019
Others	GCMS	Normal	0.00020	0.00002
Dichloromethane	/	/	0.99789	0.00019

**o) Purity table for the Nitrogen**

Compounds	Amount fraction (10 <sup>-6</sup> mol/mol)	Standard uncertainty (10 <sup>-6</sup> mol/mol)	Assumed distribution
CH <sub>4</sub>	0.001	0.0005	Rectangular
Ar	48.3	4.8	Normal
O <sub>2</sub>	0.02	0.006	Normal
CO <sub>2</sub>	0.010	0.003	Normal
H <sub>2</sub> O	0.10	0.04	Normal
N <sub>2</sub> O	0.0001	0.0004	Rectangular
N <sub>2</sub>	999952.1	4.8	Normal

**p) Purity table for the final mixture**

**(Standard 1)** Cylinder Identification Number: FF 56504

Component	Mole fraction Value	Unit	Expanded Uncertainty	Unit	Coverage Factor
CH <sub>4</sub>	9.9894E-10	mol/mol	9.376E-10	mol/mol	2
Ar	4.8249E-05	mol/mol	9.001E-06	mol/mol	2
O <sub>2</sub>	1.9979E-08	mol/mol	1.125E-08	mol/mol	2
CO <sub>2</sub>	9.9894E-09	mol/mol	5.626E-09	mol/mol	2
H <sub>2</sub> O	1.0189E-07	mol/mol	7.501E-08	mol/mol	2

N2O	9.9894E-12	mol/mol	5.438E-12	mol/mol	2
N2	9.9995E-01	mol/mol	2.115E-04	mol/mol	2
other	1.3953E-09	mol/mol	1.525E-10	mol/mol	2
C4H6	1.0511E-07	mol/mol	2.113E-04	mol/mol	2
C2HCl3	1.0482E-07	mol/mol	1.567E-10	mol/mol	2
CCl4	1.0368E-07	mol/mol	1.816E-10	mol/mol	2
CHCl3	1.0271E-07	mol/mol	1.608E-10	mol/mol	2
C2H4Cl2	1.0666E-07	mol/mol	4.155E-10	mol/mol	2
CH2Cl2	1.0518E-07	mol/mol	1.614E-10	mol/mol	2
C8H8	1.0074E-07	mol/mol	9.184E-10	mol/mol	2
C7H8	1.0263E-07	mol/mol	9.355E-10	mol/mol	2
C6H6	1.0219E-07	mol/mol	9.531E-10	mol/mol	2
C8H10(ethylbenzene)	1.0416E-07	mol/mol	9.849E-10	mol/mol	2
C8H10(m)	1.0363E-07	mol/mol	9.701E-10	mol/mol	2
C8H10(o)	1.0225E-07	mol/mol	9.638E-10	mol/mol	2
C8H10(p)	2.1566E-10	mol/mol	4.310E-11	mol/mol	2
C2Cl4	1.0269E-07	mol/mol	8.663E-10	mol/mol	2
C2H3Cl3	1.0174E-07	mol/mol	9.937E-10	mol/mol	2

**(Standard 2) Cylinder Identification Number: FF 57601**

Component	Mole fraction Value	Unit	Expanded Uncertainty	Unit	Coverage Factor
CH4	9.990E-10	mol/mol	9.375E-10	mol/mol	2
Ar	4.825E-05	mol/mol	9.000E-06	mol/mol	2
O2	1.998E-08	mol/mol	1.125E-08	mol/mol	2
CO2	9.990E-09	mol/mol	5.625E-09	mol/mol	2
H2O	1.019E-07	mol/mol	7.500E-08	mol/mol	2
N2O	9.990E-12	mol/mol	5.438E-12	mol/mol	2
N2	1.000E+00	mol/mol	2.076E-04	mol/mol	2
other	1.031E-07	mol/mol	2.073E-04	mol/mol	2
C4H6	1.443E-09	mol/mol	1.577E-10	mol/mol	2
C2HCl3	1.037E-07	mol/mol	1.554E-10	mol/mol	2
CCl4	1.026E-07	mol/mol	1.799E-10	mol/mol	2
CHCl3	1.016E-07	mol/mol	1.593E-10	mol/mol	2
C2H4Cl2	1.055E-07	mol/mol	4.112E-10	mol/mol	2
CH2Cl2	1.041E-07	mol/mol	1.600E-10	mol/mol	2
C8H8	1.041E-07	mol/mol	9.490E-10	mol/mol	2
C7H8	1.061E-07	mol/mol	9.666E-10	mol/mol	2
C6H6	1.057E-07	mol/mol	9.849E-10	mol/mol	2
C8H10	1.077E-07	mol/mol	1.018E-09	mol/mol	2
C8H10	1.071E-07	mol/mol	1.002E-09	mol/mol	2
C8H10	1.057E-07	mol/mol	9.959E-10	mol/mol	2

C8H10	2.230E-10	mol/mol	4.456E-11	mol/mol	2
C2C14	1.062E-07	mol/mol	8.951E-10	mol/mol	2
C2H3Cl3	1.052E-07	mol/mol	1.027E-09	mol/mol	2

**(Standard 3)** Cylinder Identification Number: FF 57619

Component	Mole fraction Value	Unit	Expanded Uncertainty	Unit	Coverage Factor
CH4	9.989E-10	mol/mol	9.343E-10	mol/mol	2
Ar	4.825E-05	mol/mol	8.969E-06	mol/mol	2
O2	1.998E-08	mol/mol	1.121E-08	mol/mol	2
CO2	9.989E-09	mol/mol	5.606E-09	mol/mol	2
H2O	1.019E-07	mol/mol	7.474E-08	mol/mol	2
N2O	9.989E-12	mol/mol	5.419E-12	mol/mol	2
N2	1.000E+00	mol/mol	2.172E-04	mol/mol	2
other	1.079E-07	mol/mol	2.169E-04	mol/mol	2
C4H6	1.422E-09	mol/mol	1.554E-10	mol/mol	2
C2HCl3	1.038E-07	mol/mol	1.716E-10	mol/mol	2
CCl4	1.035E-07	mol/mol	1.889E-10	mol/mol	2
CHCl3	9.713E-08	mol/mol	1.758E-10	mol/mol	2
C2H4Cl2	1.047E-07	mol/mol	4.212E-10	mol/mol	2
CH2Cl2	1.025E-07	mol/mol	2.043E-10	mol/mol	2
C8H8	1.035E-07	mol/mol	9.375E-10	mol/mol	2
C7H8	1.049E-07	mol/mol	9.509E-10	mol/mol	2
C6H6	1.034E-07	mol/mol	9.597E-10	mol/mol	2
C8H10	1.055E-07	mol/mol	9.932E-10	mol/mol	2
C8H10	1.037E-07	mol/mol	9.697E-10	mol/mol	2
C8H10	1.054E-07	mol/mol	9.865E-10	mol/mol	2
C8H10	2.157E-10	mol/mol	4.310E-11	mol/mol	2
C2C14	1.042E-07	mol/mol	8.744E-10	mol/mol	2
C2H3Cl3	1.033E-07	mol/mol	1.005E-09	mol/mol	2

**(Standard 4)** Cylinder Identification Number: FF 57609

Component	Mole fraction Value	Unit	Expanded Uncertainty	Unit	Coverage Factor
CH4	9.987E-10	mol/mol	9.221E-10	mol/mol	2
Ar	4.824E-05	mol/mol	8.852E-06	mol/mol	2
O2	1.997E-08	mol/mol	1.107E-08	mol/mol	2
CO2	9.987E-09	mol/mol	5.533E-09	mol/mol	2
H2O	1.023E-07	mol/mol	7.377E-08	mol/mol	2
N2O	9.987E-12	mol/mol	5.348E-12	mol/mol	2
N2	1.000E+00	mol/mol	2.511E-04	mol/mol	2
other	1.248E-07	mol/mol	2.509E-04	mol/mol	2
C4H6	1.680E-09	mol/mol	1.837E-10	mol/mol	2

C2HCl3	1.267E-07	mol/mol	1.808E-10	mol/mol	2
CCl4	1.263E-07	mol/mol	2.051E-10	mol/mol	2
CHCl3	1.186E-07	mol/mol	1.904E-10	mol/mol	2
C2H4Cl2	1.278E-07	mol/mol	5.030E-10	mol/mol	2
CH2Cl2	1.251E-07	mol/mol	2.264E-10	mol/mol	2
C8H8	1.223E-07	mol/mol	1.101E-09	mol/mol	2
C7H8	1.240E-07	mol/mol	1.117E-09	mol/mol	2
C6H6	1.221E-07	mol/mol	1.128E-09	mol/mol	2
C8H10	1.247E-07	mol/mol	1.167E-09	mol/mol	2
C8H10	1.225E-07	mol/mol	1.140E-09	mol/mol	2
C8H10	1.245E-07	mol/mol	1.159E-09	mol/mol	2
C8H10	2.549E-10	mol/mol	5.093E-11	mol/mol	2
C2Cl4	1.232E-07	mol/mol	1.026E-09	mol/mol	2
C2H3Cl3	1.221E-07	mol/mol	1.181E-09	mol/mol	2

**q) The repeatability and consistency of verification**

Repeatability check

One sample was injected 6 times to verify the repeatability of analytical method.

Compound	Diff.-1	Diff.-2	Diff.-3	Diff.-4	Diff.-5	Diff.-6
1,3-butadiene	0.39%	0.34%	0.14%	-0.09%	-0.21%	-0.36%
Dichloromethane	0.45%	-0.05%	-0.01%	-0.22%	-0.27%	-0.44%
1,1-dichloroethane	0.23%	-0.01%	0.19%	0.11%	0.23%	-0.04%
Chloroform	0.47%	0.10%	0.08%	-0.11%	-0.18%	-0.33%
Carbon tetrachloride	0.25%	-0.22%	0.17%	0.12%	-0.06%	-0.35%
Benzene	0.26%	0.16%	0.34%	0.07%	-0.10%	-0.27%
Trichloroethylene	-0.01%	-0.02%	0.19%	0.02%	-0.05%	-0.22%
Toluene	-0.41%	-0.42%	0.21%	0.23%	-0.04%	-0.33%
1,1,2-trichloroethane	0.26%	-0.01%	0.29%	0.06%	-0.19%	-0.31%
Tetrachloroethylene	0.13%	-0.01%	0.26%	0.09%	-0.12%	-0.29%
Ethylbenzene	0.04%	-0.28%	0.26%	0.12%	-0.22%	-0.44%
m-xylene	0.24%	-0.21%	0.41%	0.25%	-0.14%	-0.36%
o-xylene	0.01%	-0.22%	0.37%	0.07%	-0.39%	-0.43%
Styrene	0.13%	-0.53%	0.01%	0.08%	-0.43%	-0.69%

Consistency verification

Two cylinders were measured against each other to verify the consistency of the prepared standards. These cylinders are FF57601 and FF565041, of which parent gas was from different mixture.

Compound	Diff.-1	Diff.-2	Diff.-3	Diff.-4	Diff.-5	Diff.-6
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1,3-butadiene	-0.19%	-0.29%	-0.20%	-0.20%	-0.20%	-0.19%
Dichloromethane	-1.18%	-1.23%	-0.93%	-0.95%	-0.84%	-0.73%
1,1-dichloroethane	-0.10%	-0.12%	-0.08%	-0.09%	-0.11%	-0.10%
Chloroform	-0.92%	-0.83%	-1.07%	-0.66%	-0.91%	-0.52%
Carbon tetrachloride	-0.89%	-1.12%	-1.46%	-0.76%	-0.39%	-0.52%
Benzene	0.38%	0.56%	0.40%	0.45%	0.33%	0.31%
Trichloroethylene	-1.31%	-1.30%	-1.31%	-1.30%	-1.26%	-1.22%
Toluene	0.10%	0.10%	0.10%	0.15%	0.10%	0.10%
1,1,2-trichloroethane	0.01%	-0.08%	-0.35%	-0.51%	-0.46%	-0.40%
Tetrachloroethylene	-0.10%	-0.11%	-0.07%	-0.08%	-0.07%	-0.07%
Ethylbenzene	0.10%	0.08%	0.12%	0.15%	0.18%	0.19%
m-xylene	0.19%	0.16%	0.16%	0.20%	0.24%	0.14%
o-xylene	1.01%	0.75%	0.56%	0.43%	0.37%	0.21%
Styrene	-1.06%	-1.26%	-1.10%	-0.71%	-0.54%	-0.36%

#### r) Adsorption loss

The adsorption loss was determined by the following method: about 60 L gases was decanted to the target cylinder (C.A. 10 bar). The daughter cylinder was measured against mother cylinder after 30 days of stabilization.

Compound	1	2	3	4	AVE
1,3-butadiene	0.07%	0.27%	-0.13%	-0.16%	0.01%
Dichloromethane	-0.31%	0.00%	0.17%	-0.24%	-0.09%
1,1-dichloroethane	0.33%	0.35%	-0.15%	-0.56%	-0.01%
Chloroform	-0.23%	-0.16%	-0.14%	-0.38%	-0.23%
Carbon tetrachloride	-0.66%	-0.47%	-0.48%	-0.50%	-0.53%
Benzene	-0.32%	-0.07%	0.01%	-0.13%	-0.13%
Trichloroethylene	-0.65%	-0.31%	-0.20%	-0.33%	-0.37%
Toluene	-0.40%	-0.16%	-0.15%	-0.28%	-0.25%
1,1,2-trichloroethane	-0.82%	-0.59%	-0.80%	-0.88%	-0.77%
Tetrachloroethylene	-0.51%	-0.10%	-0.03%	-0.37%	-0.25%
Ethylbenzene	-0.53%	-0.24%	-0.31%	-0.51%	-0.40%
m-xylene	-0.49%	-0.22%	-0.34%	-0.72%	-0.44%
o-xylene	-0.90%	-0.63%	-0.57%	-0.76%	-0.71%
Styrene	-1.78%	-1.60%	-1.56%	-1.54%	-1.62%



Report form

Laboratory name: National Metrology Institute of South Africa

Cylinder number: D51 7565

Table 1: Measurement 1<sup>#</sup>

Component	Date (25/10/19)	Result ( $\mu\text{mol/mol}$ )	Standard deviation (% relative)	Number of replicates
1,3-butadiene	0,10699	0,00053	0,49191	Five
Dichloromethane	0,0963	0,0028	2,9250	Five
Chloroform	0,1138	0,0018	1,6150	Five
Benzene	0,1168	0,0024	2,0742	Five
Trichloroethylene	0,0996	0,0025	2,4969	Five
Toluene	0,1175	0,0029	2,4443	Five
Tetrachloroethylene	0,1284	0,0046	3,5904	Five
Ethylbenzene	0,1137	0,0020	1,7624	Five
m-xylene	0,1177	0,0019	1,600	Five
o-xylene	0,0942	0,0019	1,9806	Five

Table 2: Measurement 2<sup>#</sup>

Component	Date (12/11/19)	Result ( $\mu\text{mol/mol}$ )	Standard Uncertainty (% relative)	Number of replicates
1,3-butadiene	0,10486	0,00083	0,79308	Five
Dichloromethane	0,1058	0,0030	2,8664	Five
Chloroform	0,1042	0,0026	2,5452	Five
Benzene	0,1166	0,0043	3,6727	Five
Trichloroethylene	0,1057	0,0034	3,2035	Five
Toluene	0,1044	0,0044	4,1928	Five
Tetrachloroethylene	0,1310	0,0082	6,2756	Five
Ethylbenzene	0,1116	0,0038	3,3508	Five
m-xylene	0,1152	0,0034	2,9599	Five
o-xylene	0,0945	0,0030	3,1637	Five

Table 3: Measurement 3<sup>#</sup>

Component	Date (12/11/19)	Result ( $\mu\text{mol/mol}$ )	Standard Uncertainty (% relative)	Number of replicates
1,3-butadiene	0,10628	0,00050	0,47155	Five
Dichloromethane	0,1076	0,0028	2,5819	Five
Chloroform	0,1094	0,0019	1,7198	Five
Benzene	0,1225	0,0034	2,7852	Five
Trichloroethylene	0,1045	0,0026	2,5161	Five
Toluene	0,1030	0,0027	2,6589	Five
Tetrachloroethylene	0,1251	0,0038	3,0631	Five
Ethylbenzene	0,1124	0,0028	2,4887	Five
m-xylene	0,1164	0,0027	2,3182	Five
o-xylene	0,0928	0,0023	2,4635	Five

Table 4: Comparison sample results

Component	Result ( $\mu\text{mol/mol}$ )	Expanded uncertainty ( $\mu\text{mol/mol}$ )	Coverage factor
1,3-butadiene	0,1061	0,0013	2
Dichloromethane	0,1032	0,0058	2
Chloroform	0,1091	0,0043	2
Benzene	0,1186	0,0069	2
Trichloroethylene	0,1033	0,0057	2
Toluene	0,1083	0,0068	2
Tetrachloroethylene	0,128	0,012	2
Ethylbenzene	0,1126	0,0059	2
m-xylene	0,1164	0,0055	2
o-xylene	0,0938	0,0049	2

### Calibration standards

Please provide a brief description of the calibration standards used.

The calibration standards were prepared in two dilution steps from high pure chemicals of HAPs and BIP nitrogen gas. The high pure chemicals were purchased from Sigma Aldrich SA and the BIP nitrogen from Air Products SA. The first dilution step of HAPs was prepared using the syringe method whereby liquids were directly introduced into the cylinder with nitrogen as a diluent gas to the mole-fraction of 10  $\mu\text{mol/mol}$ . The 100 nmol/mol calibration standards were therefore diluted from the verified 10  $\mu\text{mol/mol}$  gas standard mixtures.

### Analytical method

Please provide a brief description of the instrumentation and method used for analysis

The analysis of HAPs was done using gas chromatography (GC) with two channels of Mass Spectrometer detector (MSD) and Flame Ionisation Detector (FID) connected to the Nutech pre-concentrator. The determination of HAPs in nitrogen was done using a single point calibration with a 100 µmol/mol HAPs in nitrogen PSGM. The comparison sample and the calibration standard were connected on the multi-position sampler box to the pre-concentrator which is connected to the GC. The column parameters for GC-MS are Restek (60 m x 320 µm x 1 µm) and for GC-FID are DB-624 column (30 m x 250 µm x 1,4 µm). The analytical conditions for FID are 60 °C holding time of 8 min, 80 °C holding time of 13 min at the rate of 5 °C/min, 150 °C holding time of 2min at the rate of 10 °C/min.

The Nutech pre-concentrator conditions were as follows;

Table 4: Cryogenic pre-concentration steps

CryoTrap 1	Glass beads trap operated at -100°C
CryoTrap 2	Tenax® trap operated at -20°C
CryoFocuser	Cryo focuser operated at -165°C
Volume pre-concentrated	200ml

The mole fraction for the sample was calculated using the following model equation

$$C_{unknown} = \frac{C_{known} \cdot Area_{unknown}}{Area_{known}}$$

### Uncertainty evaluation

Please provide a brief description of the evaluation of measurement uncertainty,

Considered uncertainty budget for the comparison sample is as follows;

$$u_{combined} = \sqrt{(u_{Meas1}^2 + u_{Meas2}^2 + u_{Meas3}^2)}/3$$

Table 1: Uncertainty budget associated with HAPs comparison sample values

Uncertainty budget										
Uncertainty contribution	1,3-butadiene	Dichloro methane	Chloroform	Benzene	Trichloro ethylene	Toluene	Tetrachloro ethylene	Ethylbenzene	Xylene (p,m)	o-xylene
Value										
Combined standard uncertainty (k=1)	0,00064	0,0029	0,0022	0,0035	0,0029	0,0034	0,0059	0,0029	0,0027	0,0024
Expanded uncertainty (k=2)	0,00128	0,0058	0,0043	0,0069	0,0057	0,0068	0,0117	0,0059	0,0055	0,0049
%REU	1,2	5,6	4,0	5,8	5,6	6,3	9,2	4,7	5,3	5,2

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Report

Laboratory name: Korea Research Institute of Standards and Science

Cylinder number: APEX1218423

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Measurement 1<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol/mol)	Standard deviation (% relative)	Number of replicates
Benzene	05/11/19	104.59	0.01	3
Toluene	05/11/19	105.19	0.02	3
Ethylbenzene	05/11/19	104.53	0.01	3
m-Xylene	05/11/19	103.70	0.01	3
Styrene	05/11/19	103.34	0.02	3
o-Xylene	05/11/19	104.46	0.09	3
Chloroform	05/11/19	105.93	0.03	3
1,1,2-Trichloroethane	05/11/19	101.66	0.16	3
Trichloroethylene	05/11/19	103.97	0.03	3
Tetrachloroethylene	05/11/19	101.38	0.01	3
1,1-dichloroethane	05/11/19	103.54	0.02	3
Carbontetrachloride	05/11/19	102.81	0.03	3
1,3-Butadiene	05/11/19	103.54	0.01	3
Dichloromethane	05/11/19	107.18	0.03	3

Measurement 2<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol/mol)	Standard deviation (% relative)	Number of replicates
Benzene	06/11/19	104.65	0.08	3
Toluene	06/11/19	105.25	0.06	3
Ethylbenzene	06/11/19	104.62	0.06	3
m-Xylene	06/11/19	103.68	0.14	3
Styrene	06/11/19	103.34	0.12	3
o-Xylene	06/11/19	104.49	0.14	3
Chloroform	06/11/19	105.96	0.07	3
1,1,2-Trichloroethane	06/11/19	101.55	0.05	3
Trichloroethylene	06/11/19	104.02	0.05	3
Tetrachloroethylene	06/11/19	101.48	0.05	3
1,1-dichloroethane	06/11/19	103.50	0.23	3
Carbontetrachloride	06/11/19	102.82	0.08	3
1,3-Butadiene	06/11/19	103.61	0.08	3
Dichloromethane	06/11/19	107.18	0.10	3

### Measurement 3<sup>#</sup>

Component	Date (dd/mm/yy)	Result (nmol/mol)	Standard deviation (% relative)	Number of replicates
Benzene	25/11/19	104.61	0.07	3
Toluene	25/11/19	105.17	0.07	3
Ethylbenzene	25/11/19	104.44	0.06	3
m-Xylene	25/11/19	103.60	0.06	3
Styrene	25/11/19	103.21	0.07	3
o-Xylene	25/11/19	104.21	0.03	3
Chloroform	25/11/19	105.91	0.063	3
1,1,2-Trichloroethane	25/11/19	101.46	0.09	3
Trichloroethylene	25/11/19	103.97	0.08	3
Tetrachloroethylene	25/11/19	101.36	0.06	3
1,1-dichloroethane	25/11/19	103.49	0.07	3
Carbontetrachloride	25/11/19	102.76	0.09	3
1,3-Butadiene	25/11/19	103.58	0.07	3
Dichloromethane	25/11/19	107.15	0.06	3

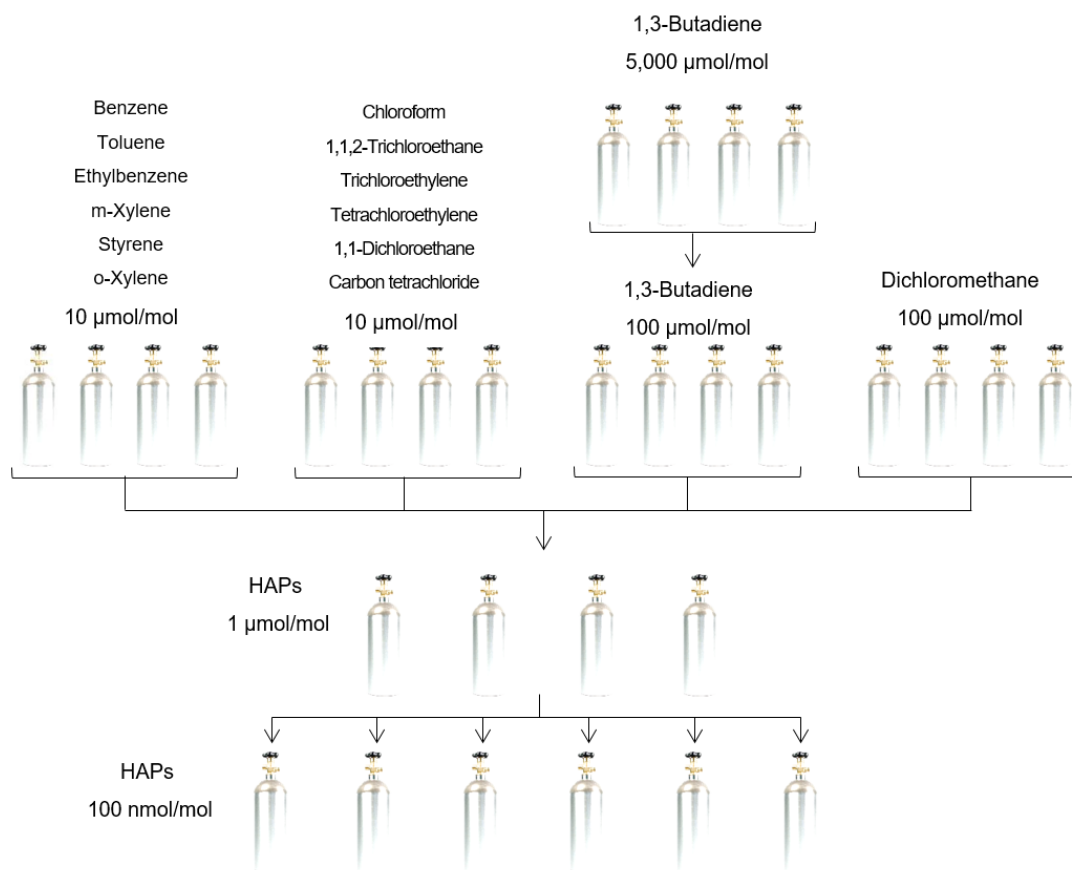
### Results

Component	Result (nmol/mol)	Expanded uncertainty (nmol/mol)	Coverage factor
Benzene	104.61	0.95	2
Toluene	105.20	0.97	2
Ethylbenzene	104.53	1.05	2
m-Xylene	103.66	1.09	2
Styrene	103.30	1.14	2
o-Xylene	104.39	1.38	2
Chloroform	105.93	1.02	2
1,1,2-Trichloroethane	101.56	1.28	2
Trichloroethylene	103.99	1.19	2
Tetrachloroethylene	101.41	1.35	2
1,1-dichloroethane	103.51	1.01	2
Carbontetrachloride	102.80	1.09	2
1,3-Butadiene	103.58	1.03	2
Dichloromethane	107.17	1.22	2

#### Calibration standards

A set of primary reference gas mixtures (PRMs) was gravimetrically prepared for the comparison. All source reagents were analyzed using GC-FID to determine their purities (based on peak areas). Micro-syringes were used to transfer the source reagents into cylinders for gravimetrically prepared PRMs at 10  $\mu\text{mol/mol}$ . The PRMs were further diluted with

nitrogen to 100 nmol/mol (Figure 1). The PRMs at each step were analyzed against each other for verification.



#### Analytical method

All analysis was carried out using GC-FID (6890, Agilent Technologies) with a cryogenic concentrator (7200 Preconcentrator, Entech Instruments). Table 1 and 2 describe the method parameters of the GC-FID and the cryogenic concentrator, respectively.

**Table 1. Method parameters for the GC-FID system**

GC-FID (Agilent 6890)	
Column	VOCOL (60m × 320 µm × 1.8 µm)
	Flow 2 mL/min (He)
Oven	105 °C (isothermal, 34.5 min)
Detector	250 °C (isothermal), H <sub>2</sub> : 35 mL/min, Air: 300 mL/min, Makeup: 15 mL/min

**Table 2. Method parameters for the cryogenic concentrator**

	Trap temperature, °C	M1→M2 temperature, °C	M2→M3 temperature, °C	Inject temperature, °C	Bake out temperature, °C
Mod 1 Trap	40	40			40

Mod 2 Trap	-95	-95	220		200
Mod 3 Trap			-170	110	
Trapping sample					
Sample flow	90 mL/min				
Sample volume	270 mL				

The KRISS and the sample mixture were analyzed using a GC-FID with a cryogenic concentrator. The analysis method consisted of six sample injections by alternating between the two cylinders (i.e., PRM<sub>ref</sub> – PRM<sub>sample</sub> – PRM<sub>ref</sub> – PRM<sub>sample</sub> – PRM<sub>ref</sub> – PRM<sub>sample</sub> – PRM<sub>ref</sub>). Please provide a brief description of the instrumentation and method used for analysis

#### Uncertainty evaluation

The measurement uncertainty consists of uncertainties from two sources such as the gravimetric preparation of the KRISS PRM and the comparison analysis. The gravimetric preparation uncertainty includes uncertainties from impurity analysis, molecular weight, weighing process, short-term stability (i.e., absorption on the internal surface of a cylinder), and internal consistency (i.e., the reproducibility of the gravimetric preparation). The analytical uncertainty is comprised of reproducibility, repeatability, and drift of GC measurements.

The amount mole fractions of the sample cylinder are determined by the following equation.

$$(1) \quad x_{sample} = x_{ref} \times R_{avg}$$

where  $x_{sample}$  is the amount mole fraction of the sample,  $x_{ref}$  is the amount mole fraction of KRISS PRM, and  $R_{avg}$  is the average of GC peak area ratios (i.e., peak area of the sample to peak area of the KRISS PRM) for nine measurements during three days.

The combined standard uncertainty is estimated as

$$\left( \frac{u(x_{sample})}{x_{sample}} \right)^2 = \left( \frac{u(x_{ref})}{x_{ref}} \right)^2 + \left( \frac{u(R_{avg})}{R_{avg}} \right)^2$$

**Table 3. Uncertainty budget for Benzene**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
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Gravimetric preparation ( $x_{ref}$ )	105.020 nmol mol <sup>-1</sup>	Normal distribution	0.442 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.004 \times x_{sample}$
Response ratio	0.996	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 0.95 nmol mol<sup>-1</sup>

**Table 4. Uncertainty budget for Toluene**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	105.551 nmol mol <sup>-1</sup>	Normal distribution	0.458 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.004 \times x_{sample}$
Response ratio	0.997	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 0.97 nmol mol<sup>-1</sup>

**Table 5. Uncertainty budget for Ethylbenzene**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	104.757 nmol mol <sup>-1</sup>	Normal distribution	0.496 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.998	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.05 nmol mol<sup>-1</sup>

**Table 6. Uncertainty budget for m-Xylene**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	103.915 nmol mol <sup>-1</sup>	Normal distribution	0.497 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$

Response ratio	0.998	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$
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Coverage factor: 2

Expanded uncertainty: 1.09 nmol mol<sup>-1</sup>

**Table 7. Uncertainty budget for Styrene**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	103.535 nmol mol <sup>-1</sup>	Normal distribution	0.525 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.998	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.14 nmol mol<sup>-1</sup>

**Table 8. Uncertainty budget for o-Xylene**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	104.407 nmol mol <sup>-1</sup>	Normal distribution	0.624 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.006 \times x_{sample}$
Response ratio	1.000	Normal distribution	0.003	$x_{sample}/R_{ref}$	$0.003 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.38 nmol mol<sup>-1</sup>

**Table 9. Uncertainty budget for Chloroform**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	107.217 nmol mol <sup>-1</sup>	Normal distribution	0.483 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.988	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.02 nmol mol<sup>-1</sup>

**Table 10. Uncertainty budget for 1,1,2-Trichloroethane**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	102.621 nmol mol <sup>-1</sup>	Normal distribution	0.613 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.006 \times x_{sample}$
Response ratio	0.990	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.28 nmol mol<sup>-1</sup>

**Table 11. Uncertainty budget for Trichloroethylene**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	105.221 nmol mol <sup>-1</sup>	Normal distribution	0.574 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.988	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.19 nmol mol<sup>-1</sup>

**Table 12. Uncertainty budget for Tetrachloroethylene**

Uncertainty source $X_i$	Estimate $x_i$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_i$	Contribution to standard uncertainty $u_i(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	102.587 nmol mol <sup>-1</sup>	Normal distribution	0.660 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.006 \times x_{sample}$
Response ratio	0.988	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.35 nmol mol<sup>-1</sup>

**Table 13. Uncertainty budget for 1,1-Dichloroethane**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	104.622 nmol mol <sup>-1</sup>	Normal distribution	0.471 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.989	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.01 nmol mol<sup>-1</sup>**Table 14. Uncertainty budget for Carbontetrachloride**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	104.059 nmol mol <sup>-1</sup>	Normal distribution	0.493 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.988	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.09 nmol mol<sup>-1</sup>**Table 15. Uncertainty budget for 1,3-Butadiene**

Uncertainty source $X_I$	Estimate $x_I$	Assumed distribution	Standard uncertainty $u(x_i)$	Sensitivity coefficient $c_I$	Contribution to standard uncertainty $u_I(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	109.896 nmol mol <sup>-1</sup>	Normal distribution	0.523 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.942	Normal distribution	0.001	$x_{sample}/R_{ref}$	$0.001 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.03 nmol mol<sup>-1</sup>**Table 16. Uncertainty budget for Dichloromethane**

Uncertainty source	Estimate $x_I$	Assumed distribution	Standard uncertainty	Sensitivity coefficient	Contribution to standard
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$X_I$			$u(x_i)$	$c_I$	uncertainty $u(y)$ , nmol mol <sup>-1</sup>
Gravimetric preparation ( $x_{ref}$ )	107.906 nmol mol <sup>-1</sup>	Normal distribution	0.589 nmol mol <sup>-1</sup>	$x_{sample}/x_{ref}$	$0.005 \times x_{sample}$
Response ratio	0.993	Normal distribution	0.002	$x_{sample}/R_{ref}$	$0.002 \times x_{sample}$

Coverage factor: 2

Expanded uncertainty: 1.22 nmol mol<sup>-1</sup>